Knowledge of Real Fluid Behaviour – the Key to Successful Gas Processing Systems

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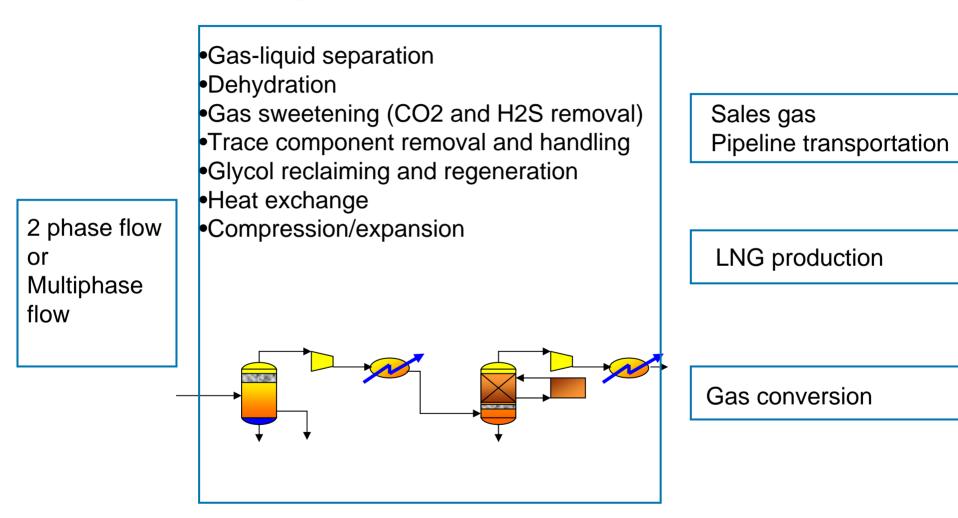


## Outline

- Definition Gas Processing
- Some Challenges Gas Processing
- Gas-Liquid Separation
- Solubility challenges
  - -Glycol Solubility in Gas
  - -Salt Solubility in Glycol



## **Gas Processing - elements**





## **Some Challenges**



#### **Gas-liquid separation**

- Compressor breakdown
- •Upsets in contactors and absorbers
- •Low efficiency/malfunction of adsorbers and absorbers

•Off-spec gas product



#### **Absorption processes**

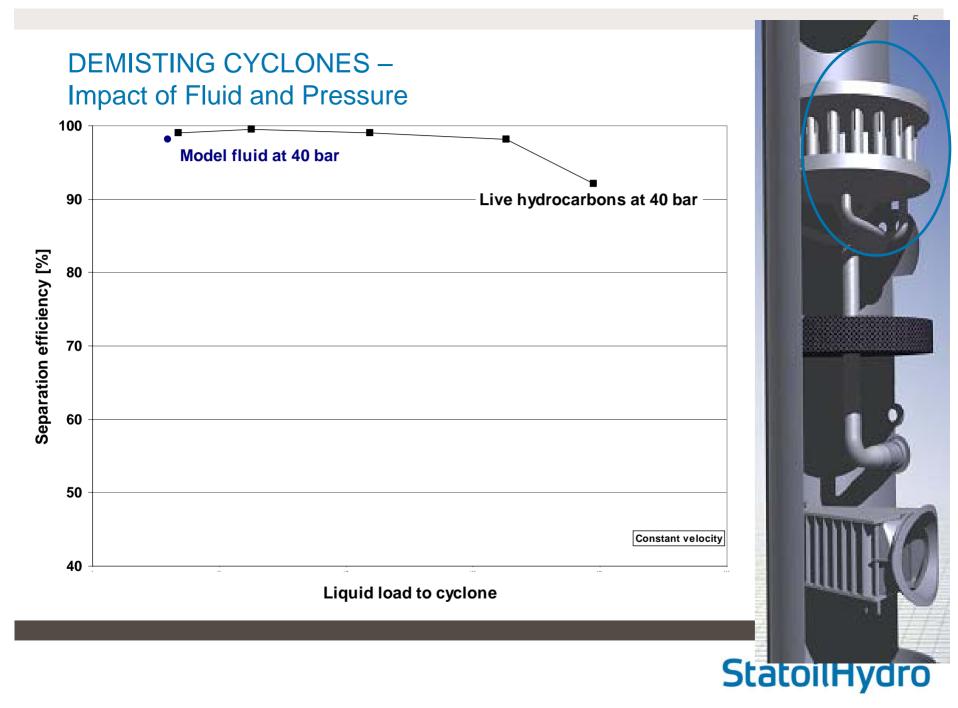
- Absorbent capacity and kinetics
- •Foaming
- •Emulsions due to additives
- Loss of absorbent



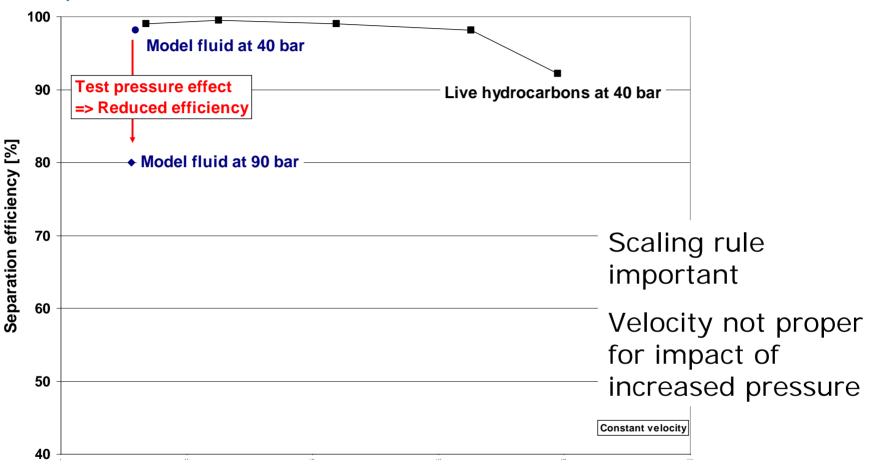
#### Solubility of trace components

- •Changes with process parameters
- •Accumulates and deposits





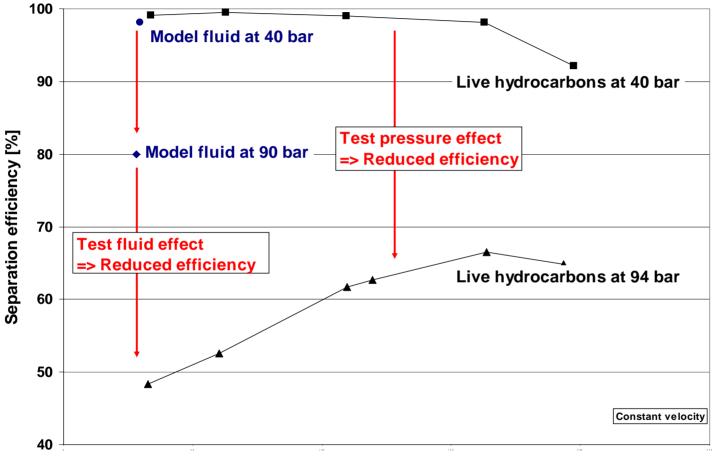
### DEMISTING CYCLONES – Impact of Fluid and Pressure



Liquid load to cyclone



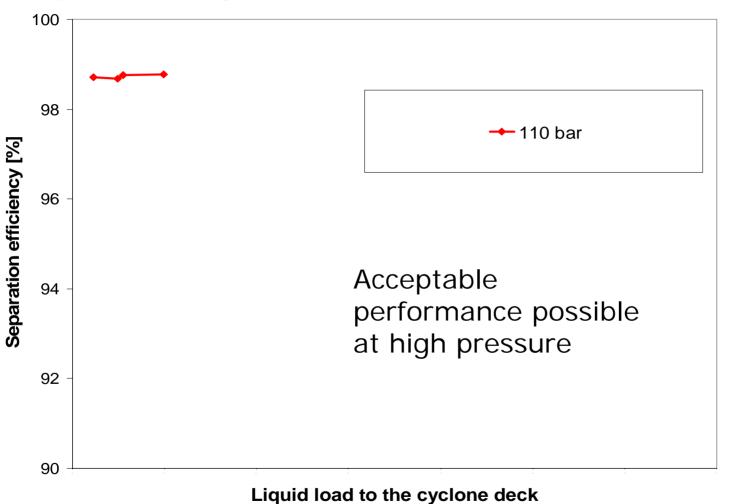
### DEMISTING CYCLONES – Impact of Fluid and Pressure

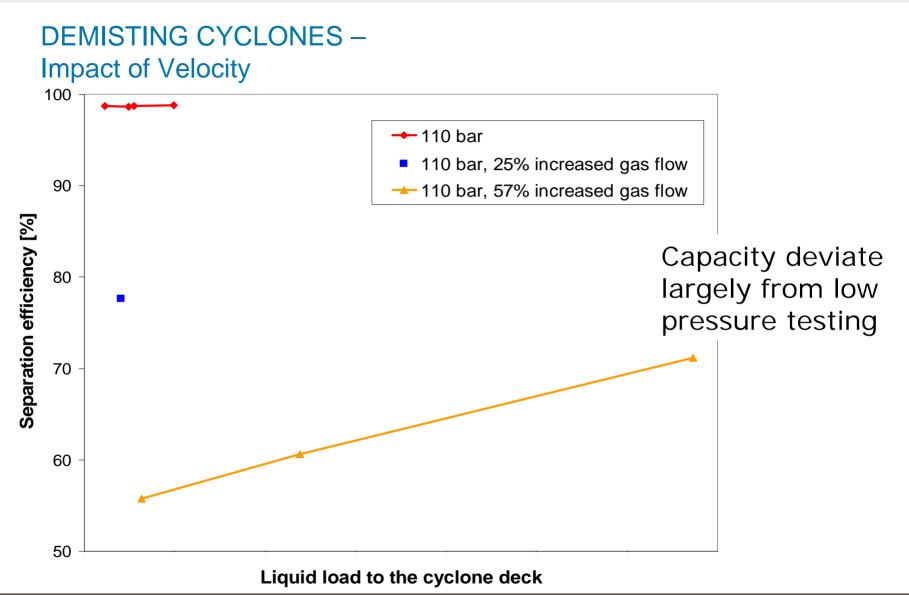


Liquid load to cyclone [l/hr]



### DEMISTING CYCLONES – Impact of Velocity





Fluid Behaviour – Gas–Liquid Separation

### Fluid impact

- Water not representative for hydrocarbon systems
- Large impact of fluid properties such as surface tension

#### • High pressure separation

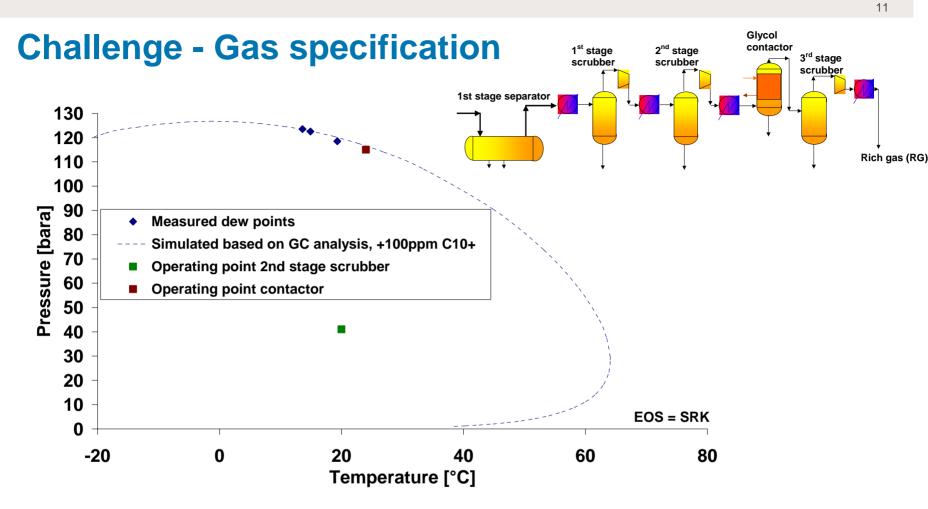
Large impact of pressure

### Scrubber elements

-Large variation in characteristics

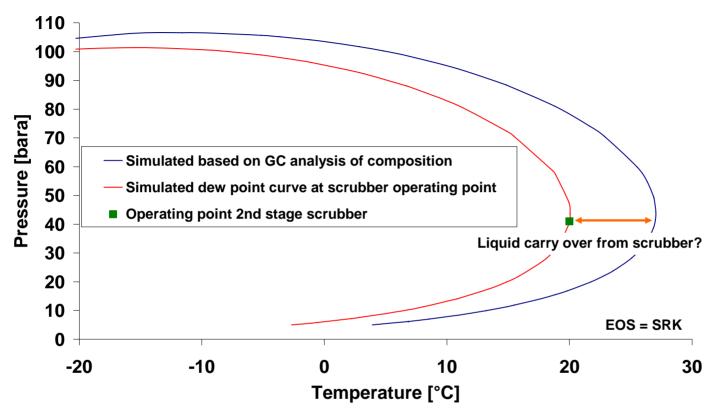
Testing: basis for fundamental understanding and establishment of proper scaling rules



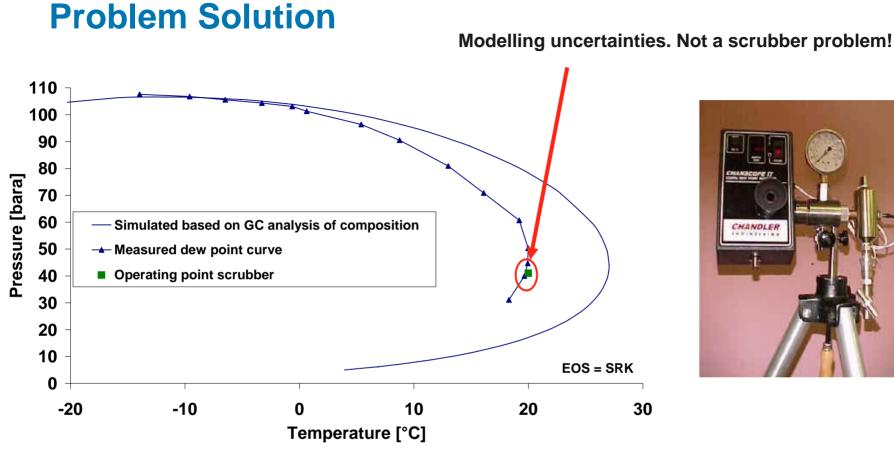


- High cricondenbar, 127 barg. Contactor determines cricondenbar
- Liquid entrainment from 2<sup>nd</sup> stage scrubber.

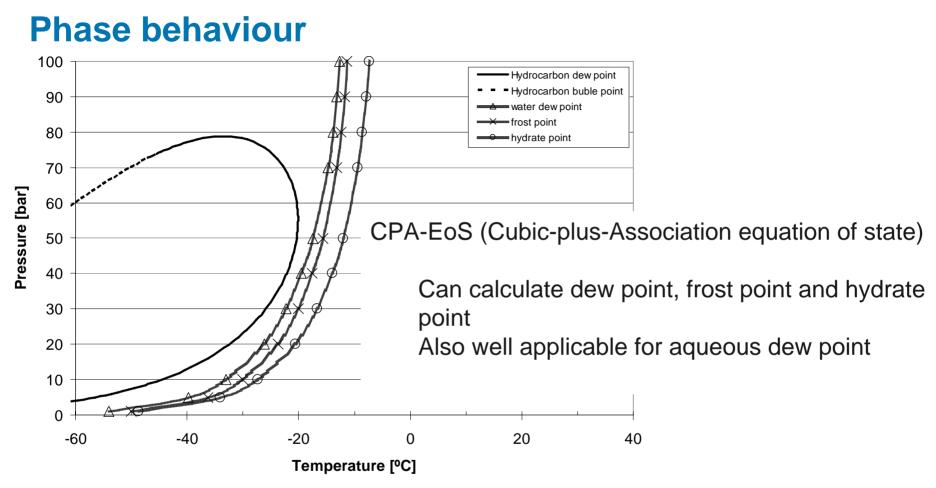
## After modification of scrubbers



- Simulated dew point curve at higher temperature than scrubber operating temperature indicates liquid carry over from scrubber
- Need for dew point measurements offshore or representative gas samples taken in single phase flow for dew point measurements in laboratory

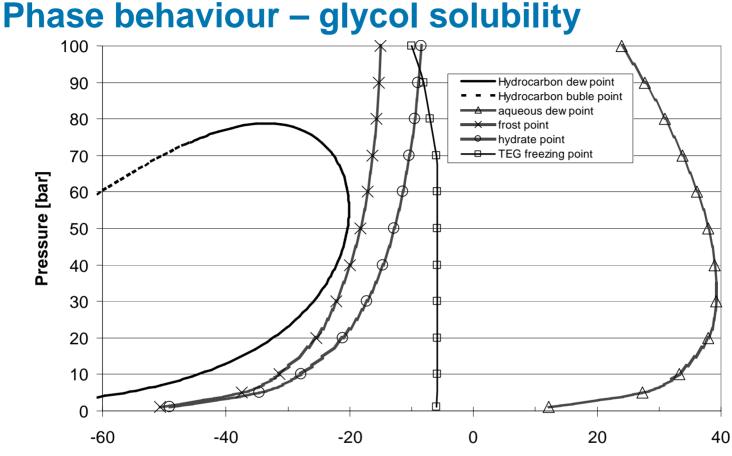


- Experimental dew point measurements needed to find correct conclusions
- True dew point curve steeper than simulated as also found for the synthetic gases



Phase behaviour of natural gas with traces of water (40 ppm(mole)),

NG composition (mole): 85 % C1, 10 % C2, 4 % C3, 0.5 % nC4, 0.5 % iC4



Temperature [°C]

Phase behaviour of natural gas with traces of water (40 ppm(mole)) and TEG (0.5 ppm(mole)),

NG composition (mole): 85 % C1, 10 % C2, 4 % C3, 0.5 % nC4, 0.5 % iC4

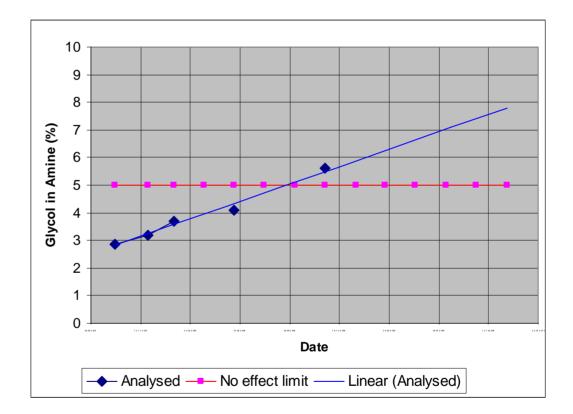
## **Challenge – Contamination of absorbent**

Glycol dehydration unit in upstream process

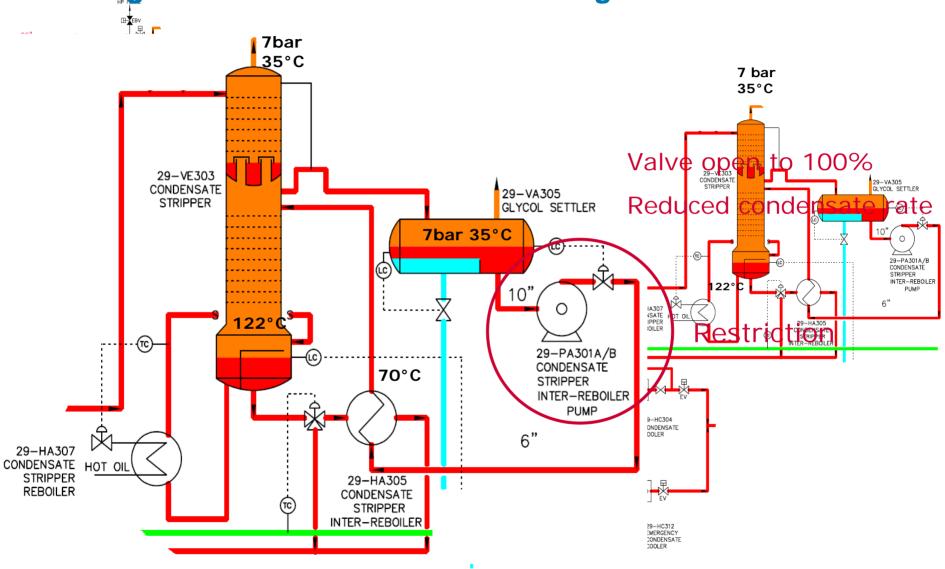
Glycol soluble in gas

Accumulates in Amine

Amine capacity affected







### Scaling in Kollsnes condensate handling

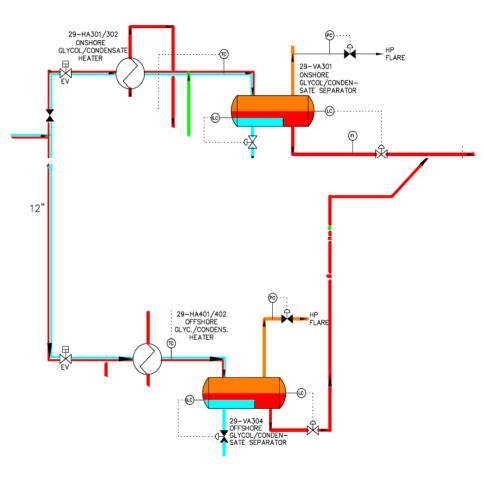
## **Questions to task force**

- Why scale in a condensate system?
- Why now after several years of operation?
- Why NaHCO<sub>3</sub> a salt with high solubility?
- How to remove it without shutting down the production?
  - A wash/replace will require 8-12 hours -> loss of 50-70 MSm<sup>3</sup> gas
- Which chemicals can we use that will not contaminate the condensate?
  And please hurry!
  The valve is about to get plugged once more!



## **Process analysis**

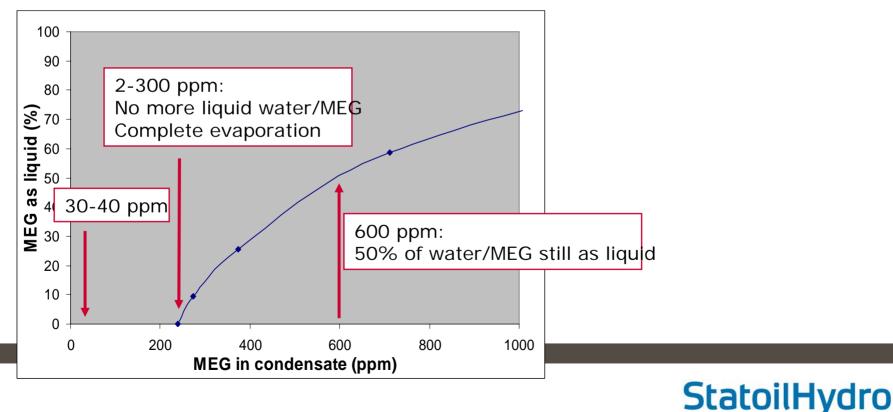
- Only one condensate MEG separator
   MEG in condensate: 600 ppm
- Start of second cond-MEG separator
  MEG in condensate: 30-40 ppm
- Improved separation, why problem?
  - Less MEG should give less salt and less precipitation?





## **MEG evaporation in stripper column**

- MEG is depressurised to 7 bar and heated to 35°C
  - Solubility of MEG and water in gas increases
- What happens when MEG in condensate is reduced from 600 to 30-40 ppm?



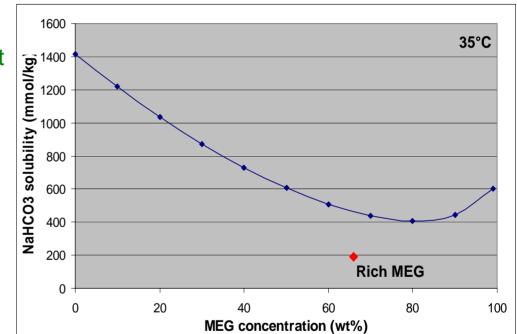
## **Scale removal options**

- Open/replace valve
  - Require shutdown, loss of 50-70 MSm<sup>3</sup> gas
- Carbonate salt -> Use an acid (suggested by a service company)
  - Require shutdown as acid would contaminate condensate
- Water NaHCO<sub>3</sub> is highly soluble
  - Possible, but will increase water content in condensate
  - May cause hydrate formation in condensate transfer line
- What about using MEG?



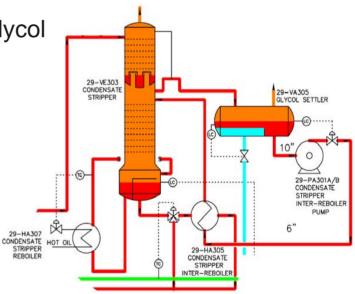
## Use of MEG to dissolve scale

- NaHCO<sub>3</sub> is soluble in MEG
- Advantages with MEG
  - Already present and will not contaminate the condensate
  - -MEG is available
  - -No use of other chemicals
  - Spent MEG can be treated in MEG regeneration



## **Treatment – MEG injection**

- MEG injected into condensate upstream stripper
- Injection rate: 40 litre/hour
  - Rate adjusted to get accumulation in glycol settler
- Treatment duration: 24 hours
  - Total MEG consumption is about 1 m<sup>3</sup>
- MEG collected in glycol settler and sent to MEG regeneration



## Summary – key to success

- Experimental evaluations
  - Evaluations have to be carried out with real fluid systems model systems will deviate from real systems
  - Large impact of pressure; high pressure processing is a challenge
  - Establish fundamental data and knowledge of mechanisms
  - Developing improved design and solutions
- Modelling
  - Experimental data and experiences need to be incorporated into models
  - Models to be used in combination with best practices
- Operational experience and problem definition
  - Important to identify where data/knowledge is needed
  - Combination with experimental experience proven to be successful