

# **CO<sub>2</sub> capture from IGCC by low-temperature syngas separation and partial condensation of CO<sub>2</sub>**

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SINTEF Energy Research

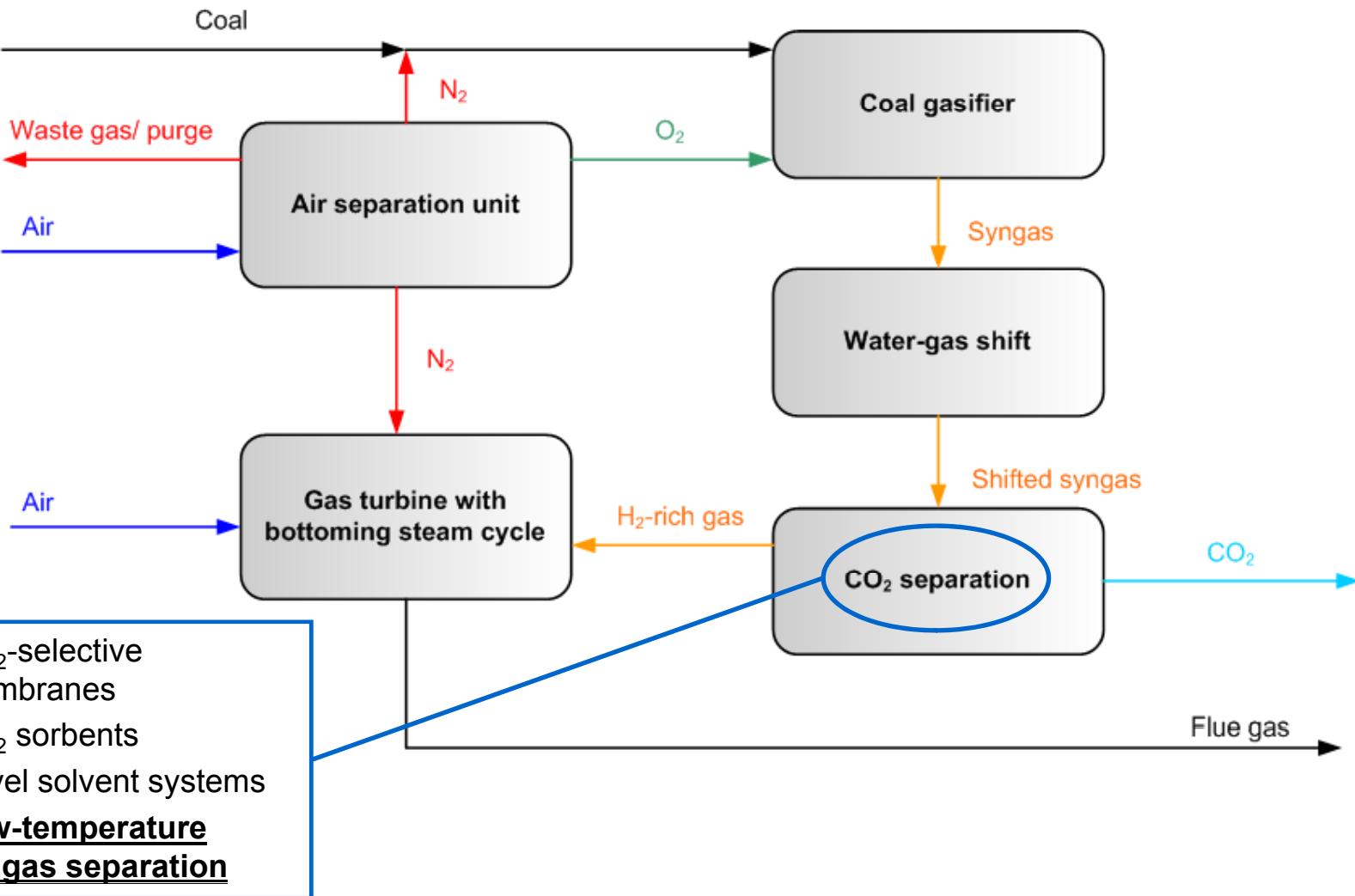
# Presentation outline

- Brief introduction to the DECARBit project
- Vapour-liquid equilibria in H<sub>2</sub>–CO<sub>2</sub> systems
  - Gives the expected CO<sub>2</sub> capture ratio for phase separation
- Principal process-level design for low-temperature syngas separation and CO<sub>2</sub> capture
- Main results from process simulations
- Conclusions

# The DECARBit project (2008–2011)

- Assess and research new techniques for **pre-combustion CO<sub>2</sub> capture**
- Develop advanced oxygen production techniques
- Continue the development efforts in FP6 projects in the pre-combustion area for key enabling technologies
- Underpin the cost reduction objective
- Establish collaborative schemes with emerging large-scale CCS initiatives in Europe
- Perform an assessment of the advanced pre-combustion capture techniques to the benefit of other energy intensive industries

# The DECARBit project (2008–2011)

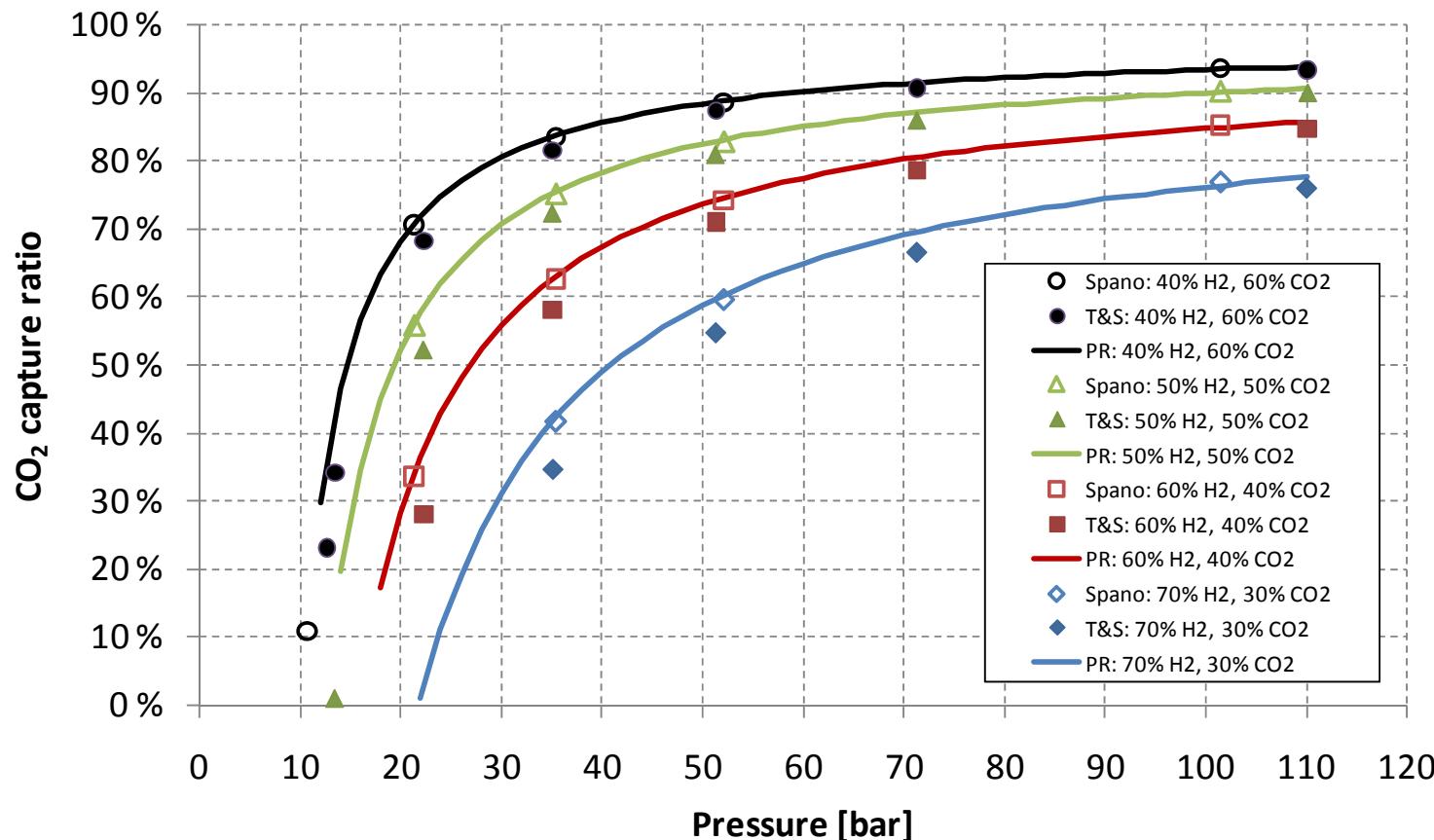


# The DECARBit project (2008–2011)

	Unit	Syngas after H <sub>2</sub> S removal, sweet shift and H <sub>2</sub> O removal
<b>Temperature</b>	°C	30
<b>Pressure</b>	bar	35.0
<b>Flowrate</b>	kg/s	114
<b>Composition</b>	mol-%	
H <sub>2</sub>		54.14
CO		1.73
CO <sub>2</sub>		38.39
N <sub>2</sub>		4.79
Ar		0.94
H <sub>2</sub> O		ppm levels
H <sub>2</sub> S		ppm levels
Other		0.02

# $\text{H}_2\text{-CO}_2$ vapour-liquid equilibria

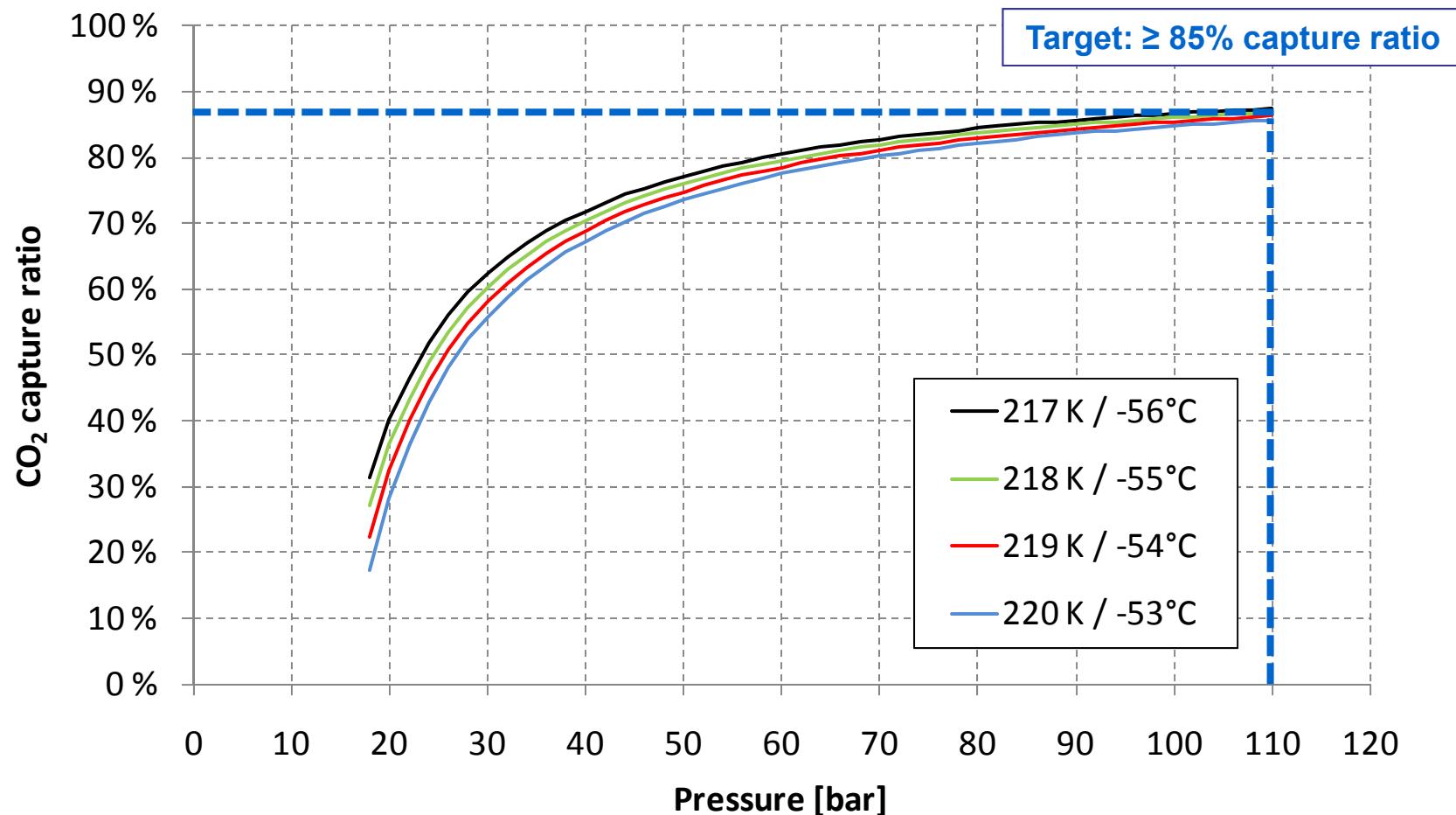
Binary  $\text{H}_2\text{-CO}_2$  mixtures at 220 K / -53°C



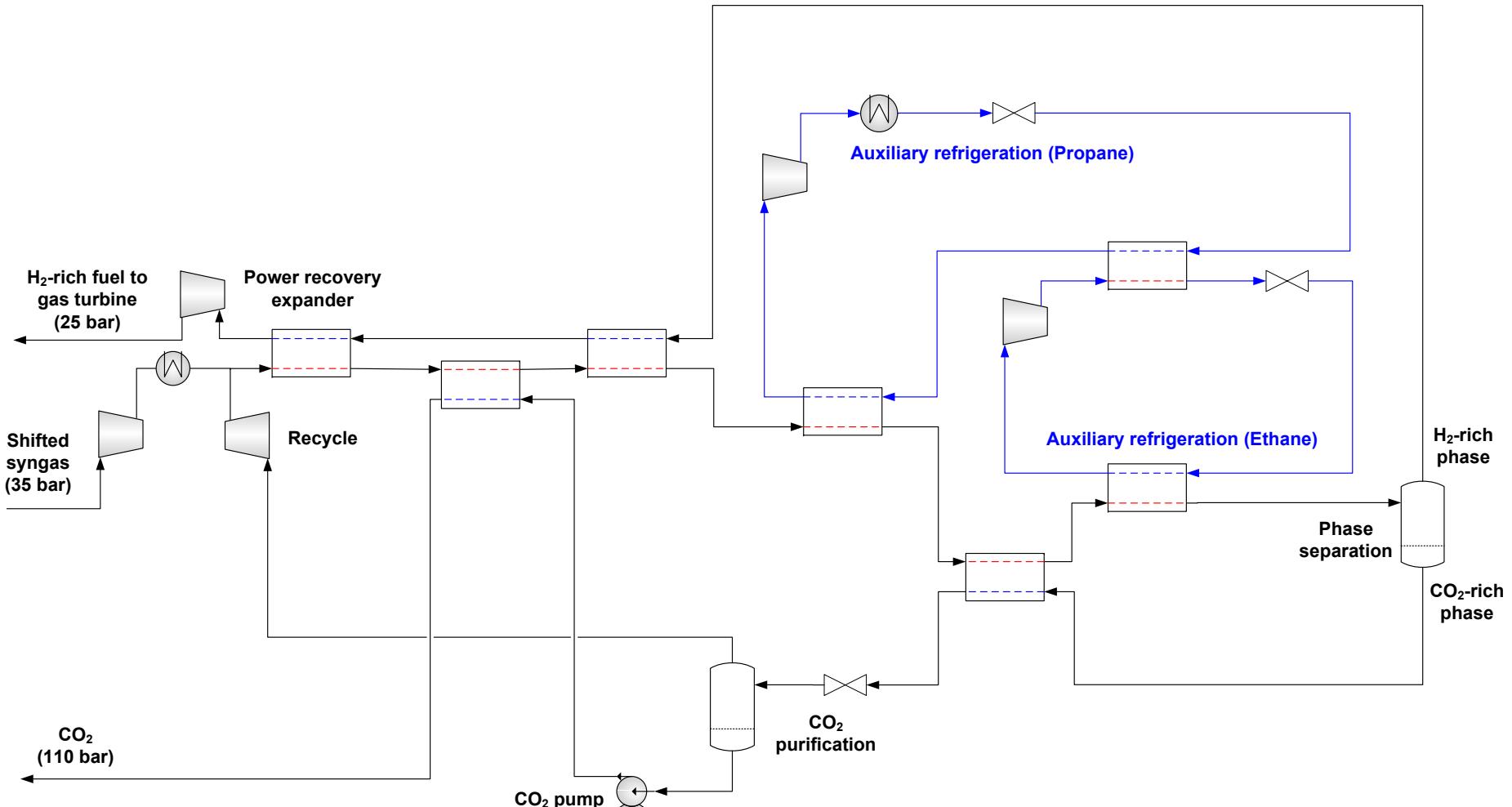
Spano J, Heck C, Barrick P. Liquid-vapor equilibria of the hydrogen–carbon dioxide system. *J. Chem. Eng. Data.* 13(2), 169–171 (1968).  
Tsang C, Streett W. Phase equilibria in the  $\text{H}_2/\text{CO}_2$  system at temperatures from 220 to 290 K and pressures to 172 MPa, *Chem. Eng. Sci.* 36, 993–1000 (1981).

# Vapour-liquid equilibria

60% H<sub>2</sub> – 40% CO<sub>2</sub> mixture



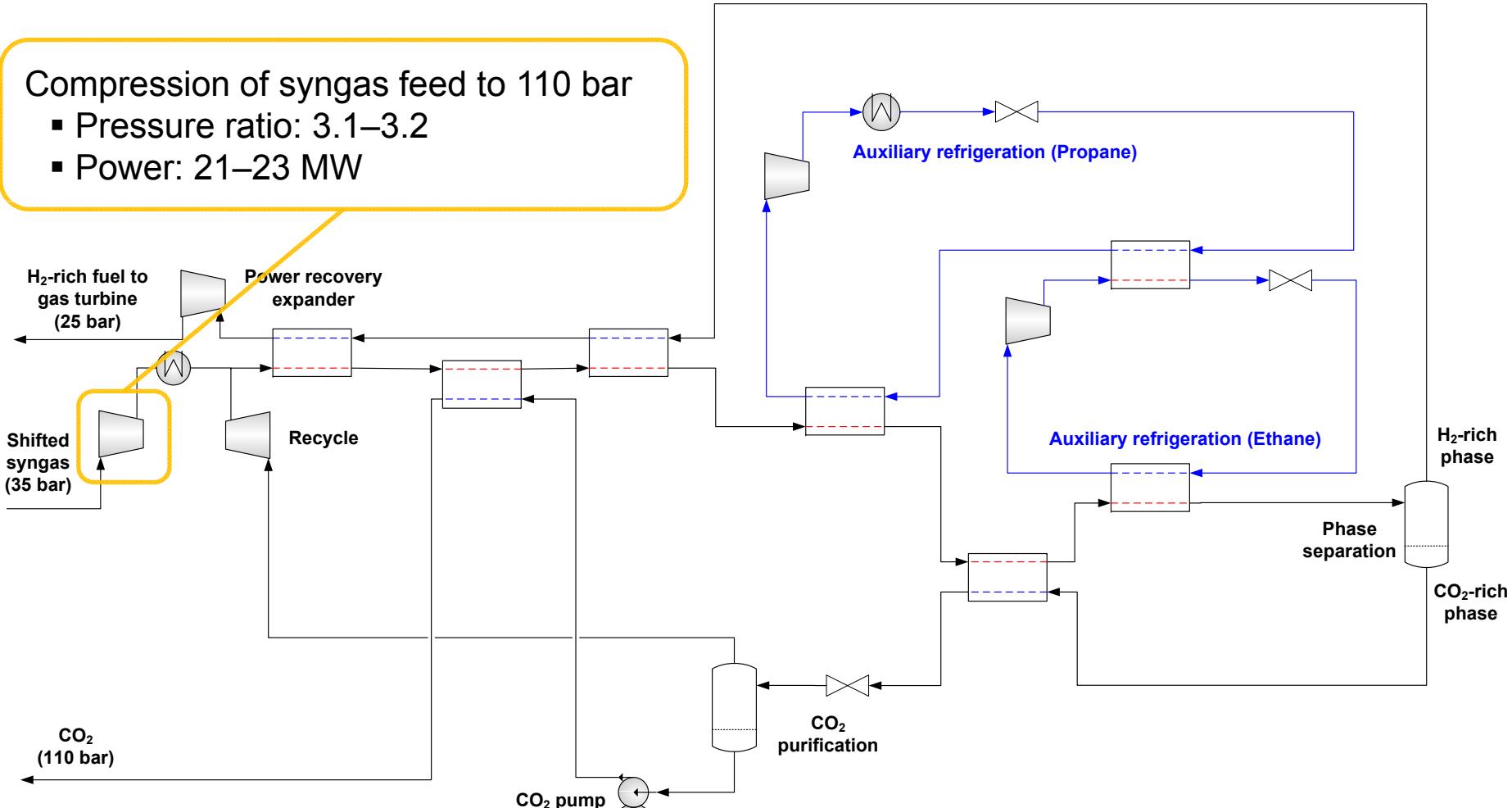
# Process flow diagram



# Process flow diagram

Compression of syngas feed to 110 bar

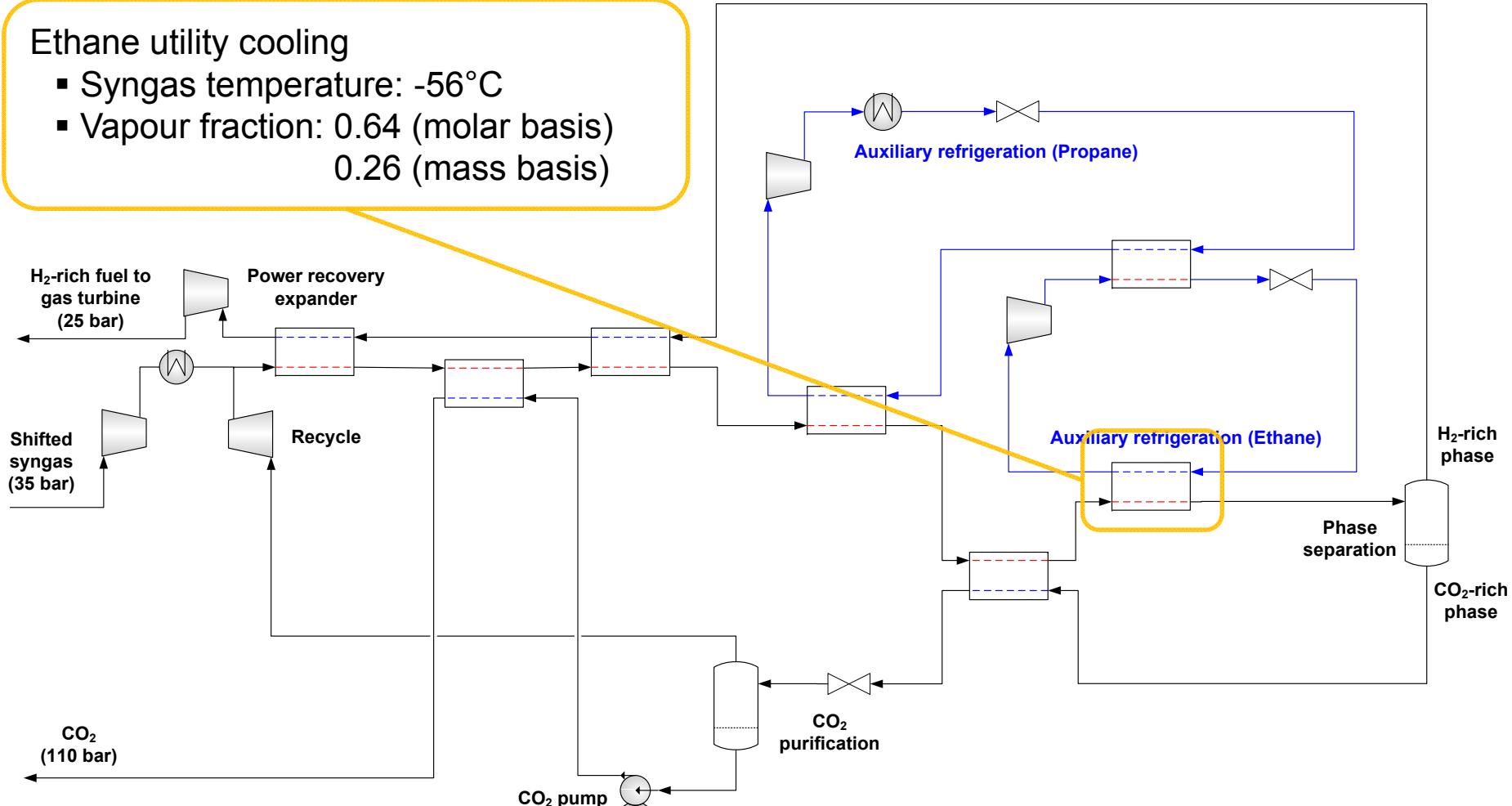
- Pressure ratio: 3.1–3.2
- Power: 21–23 MW



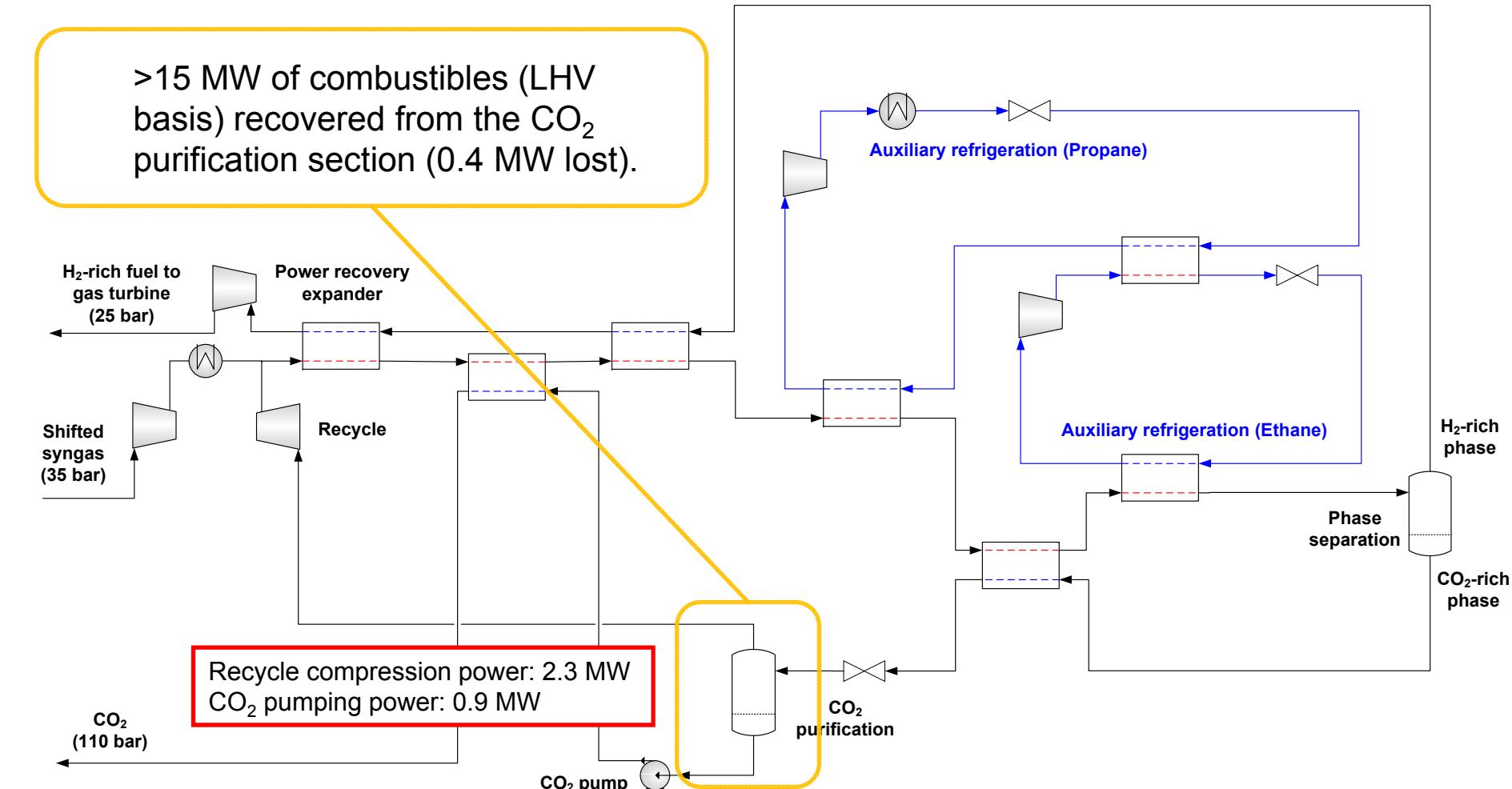
# Process flow diagram

## Ethane utility cooling

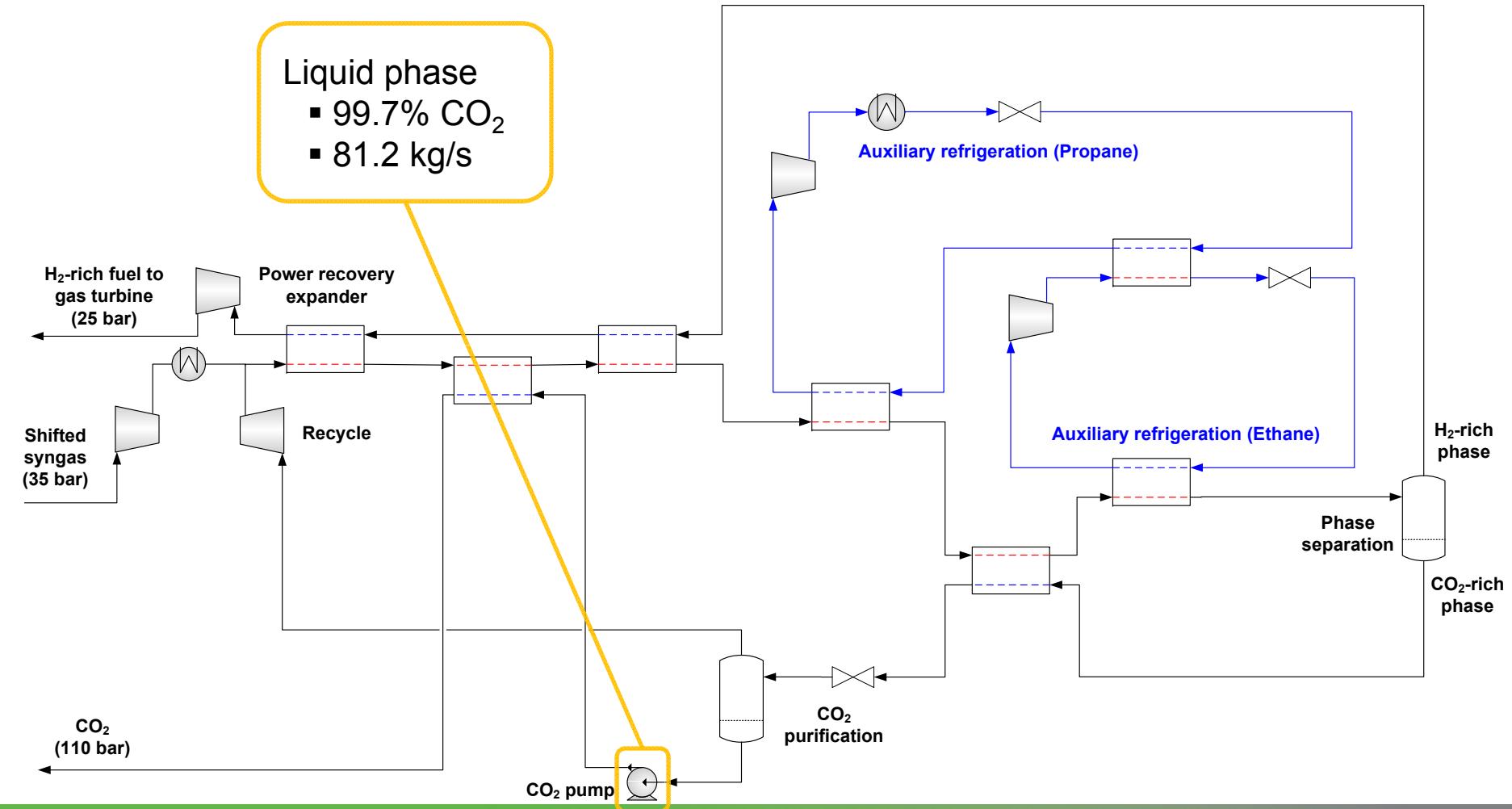
- Syngas temperature: -56°C
- Vapour fraction: 0.64 (molar basis)  
0.26 (mass basis)



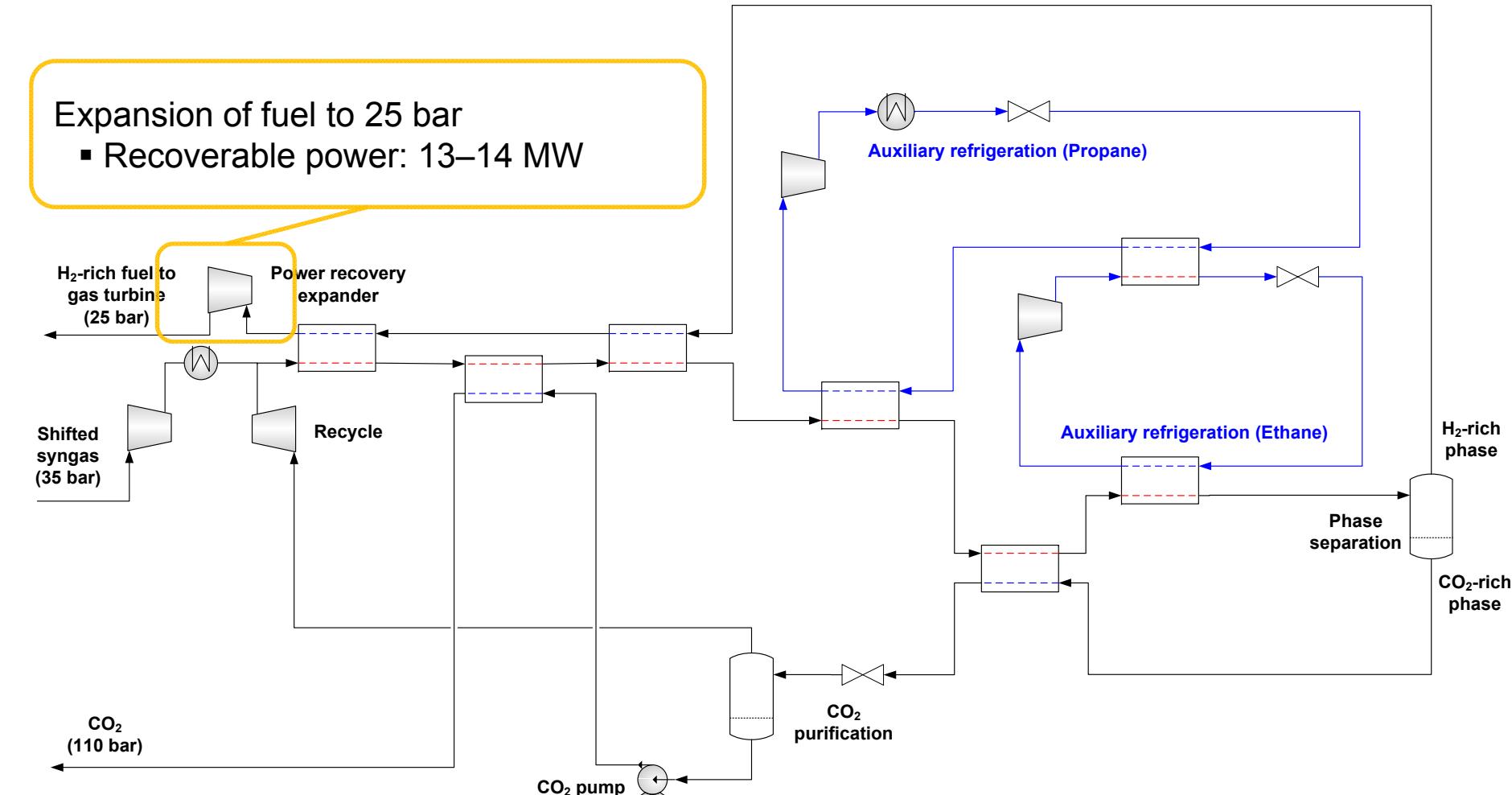
# Process flow diagram



# Process flow diagram

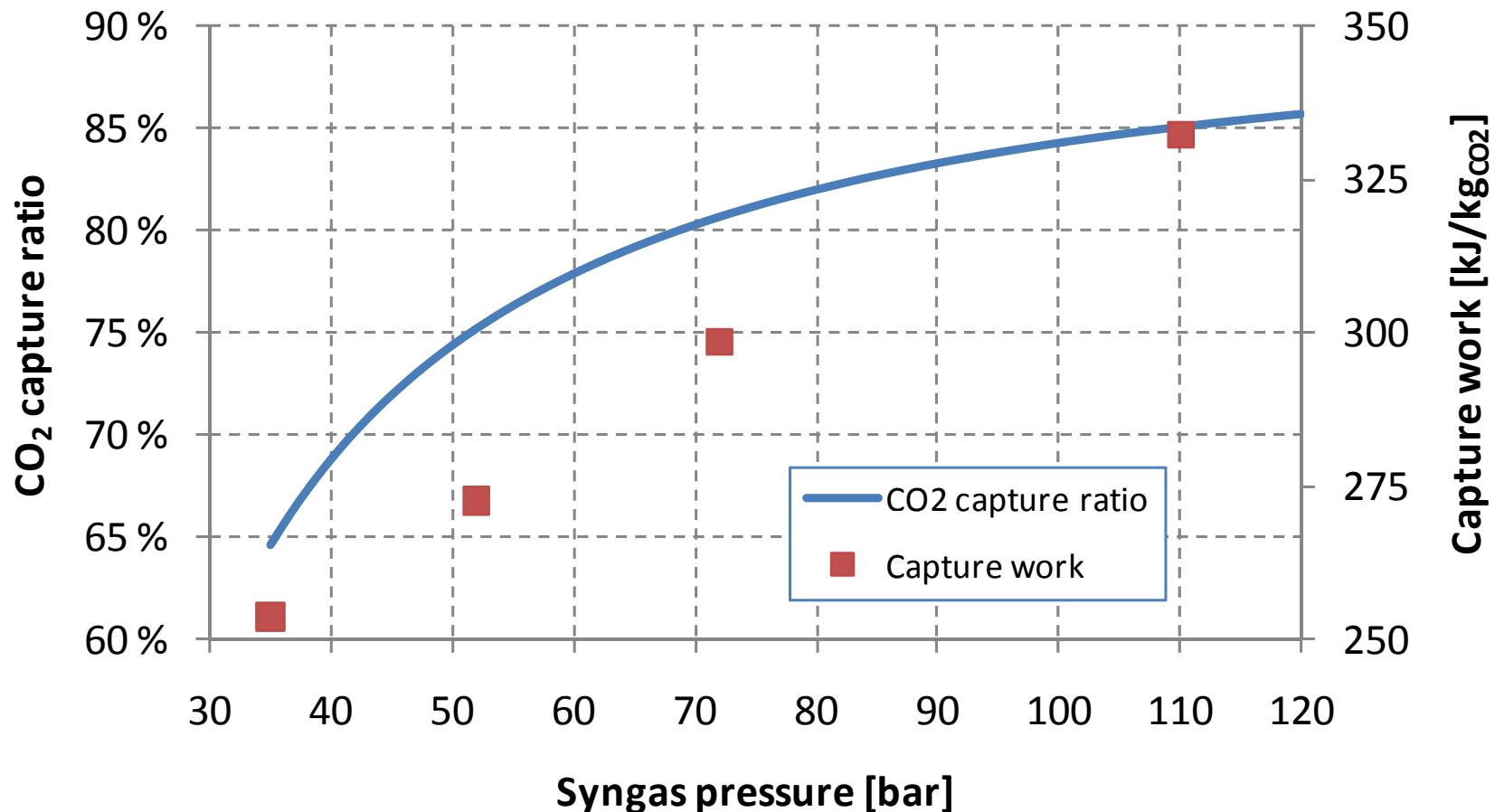


# Process flow diagram



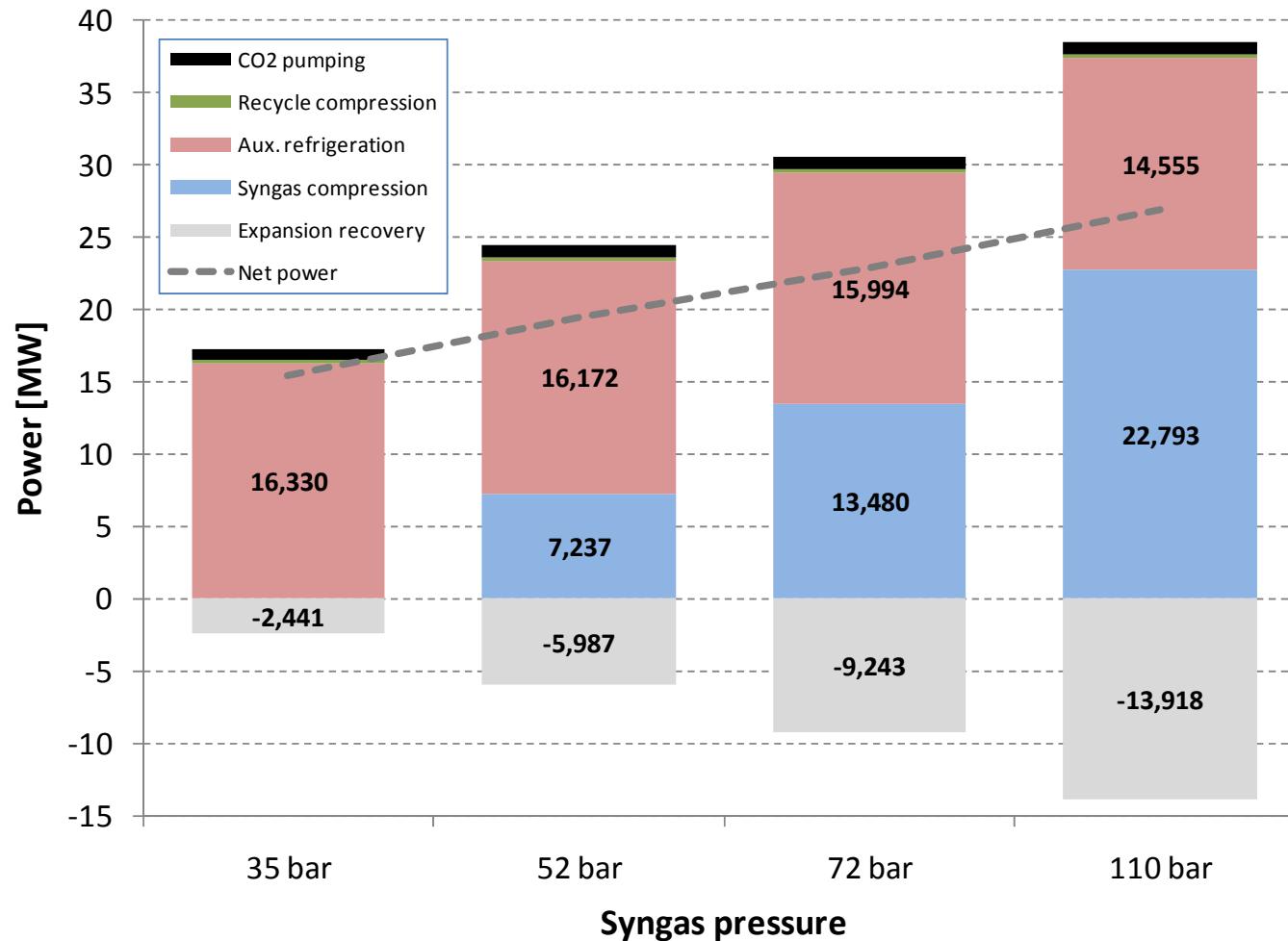
# Main results

CCR and specific capture work



# Main results

## Power consumption decomposed



# Main results

## Simulation parameters

Isentropic efficiency		
Syngas compressor	%	82
Propane compressor	%	82
Ethane compressor	%	82
Recycle compressor	%	80
Power recovery fuel expander	%	85
Liquid CO <sub>2</sub> pump	%	80
Pressure drop		
Heat exchangers	bar	0.2
Inter- and after-coolers	bar	0.5
Temperature approach		
Heat exchanger pinch	°C	3
LMTD in propane–ethane cascade heat exchanger	°C	> 5

# Concluding remarks

- A conceptual low-temperature syngas separation process for CO<sub>2</sub> capture from IGCC has been developed and simulated
- 85% capture ratio is achievable for shifted syngas with a CO<sub>2</sub> concentration of 38 mol-%. Specific capture work for the considered system boundaries is around 330 kJ per kg CO<sub>2</sub> captured
- Partial capture is possible without the need of pre-compression
- Refrigeration cycles account for a significant part of the power consumption and must be optimised
- Simulation of the overall IGCC with CO<sub>2</sub> capture is required in order to obtain a justified benchmarking with baseline technologies
  - Syngas conditioning (water removal and desulphurisation) must be included
- Overall performance analysis and techno-economic assessment, equipment sizing and costing will be carried out in the last phase of DECARBit

# Acknowledgements



Participating industry partners



Funding industry partners

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