

Institute of Combustion and Power Plant Technology

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Calcium looping for CO₂ capture in cement plants — pilot scale test

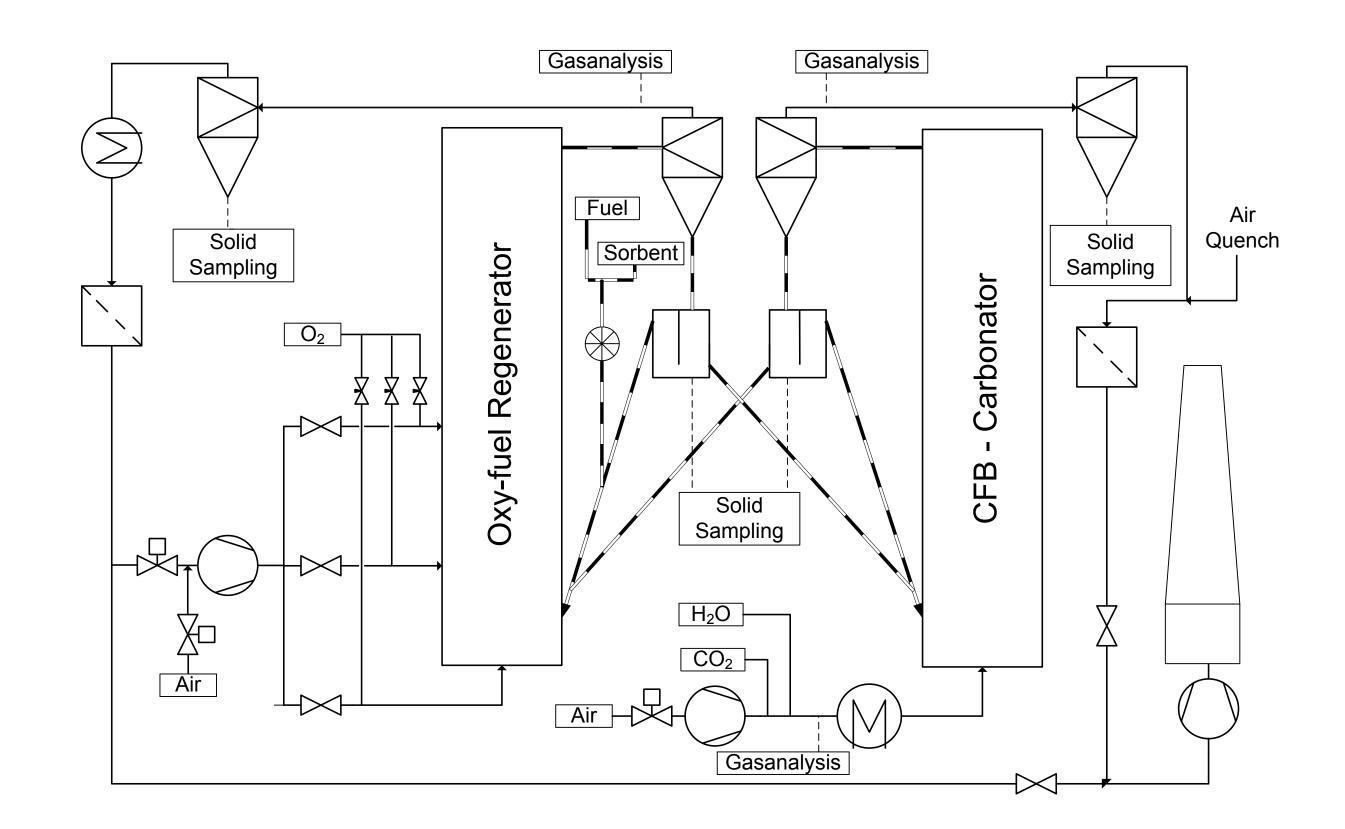
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Introduction

- Cement industry is responsible for approx. 5% of global anthropogenic CO₂ emissions
- CCS/CCU technologies shall reduce CO₂ emissions to mitigate climate change
- Calcium looping is a promising post combustion CCS technology for cement industry

Experimental Setup

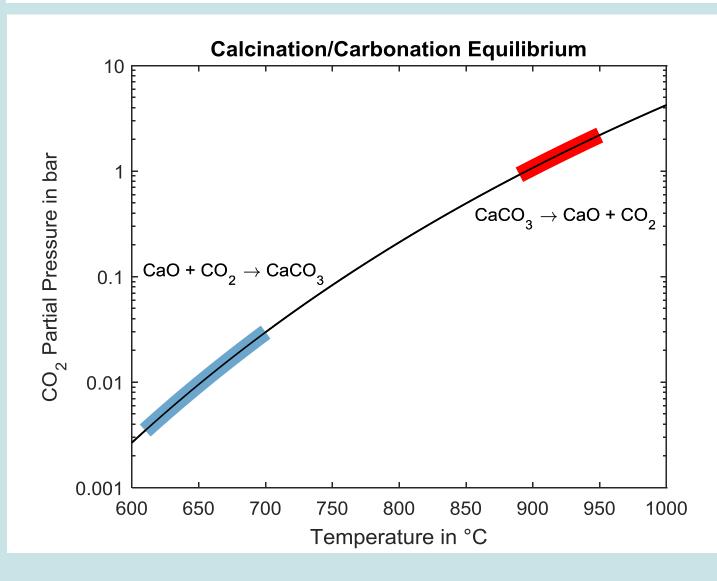
CFB-CFB configuration of 200 kW_{th} pilot plant (IFK, University of Stuttgart)



Calcium Looping

- Calcium looping is based on the cyclic calcination and carbonation of CaCO₃
- Oxy-fuel calcination leads to CO₂-rich exhaust stream ready for sequestration or utilization
- Supply of high quality energy suitable for efficient power generation or heat integration
- Low CO₂ avoiding costs due to energy integration
- Spent calcium looping sorbent can be (re)utilized in the cement process reducing its original CO₂ flue gas concentration

Calcium Looping Operation Conditions



- Oxy-fuel calcination of CaCO₃ requires temperatures around 900 °C
- Compromise between reaction kinetics and equilibrium partial pressure leads to carbonation temperature around 650 °C
- Efficient recuperation of required heat due to high temperature level
- High specific amount of fresh sorbent due to utilization in cement production
- Refractory lined interconnected circulating fluidized bed reactors (h = 10 m, d_i = 20 cm)
- Oxy-fuel calcination with recycled flue gas (wet)
- Synthetic flue gas mixture (CO_2 , H_2O , air) is fed to the carbonator
- Transfer flow controlled by cone valves
- Online gas measurement: CO₂, CO, NO, NO₂, SO₂, H₂, H₂O

Experimental Conditions

• Calcination temperature: $T_{Calc} = 890 \dots 930 \ ^{\circ}C$

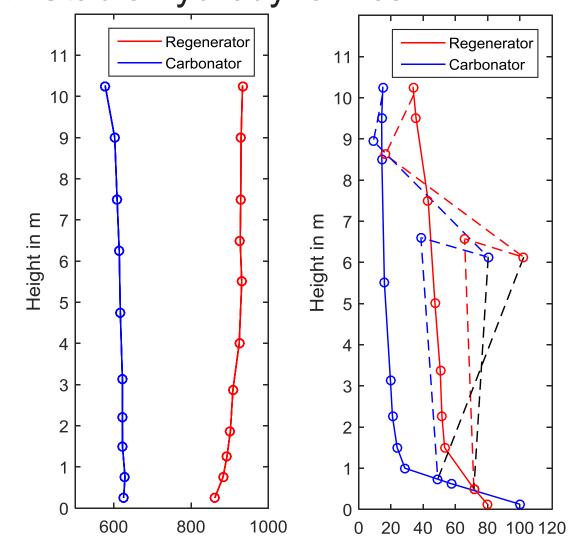
Results

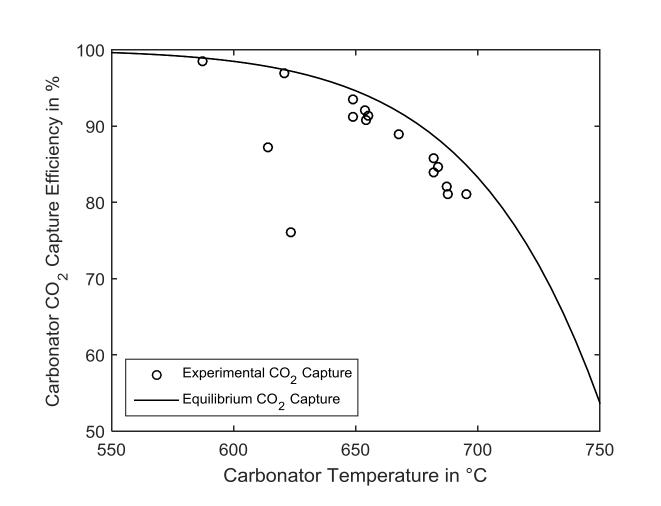
CO₂ capture performance

- CO₂ capture rates above 90 % for carbonator temperature below 650 °C and sufficient space time / looping ratio
- CO₂ capture rates near equilibrium capture rates achieved

Temperature and hydrodynamic profiles

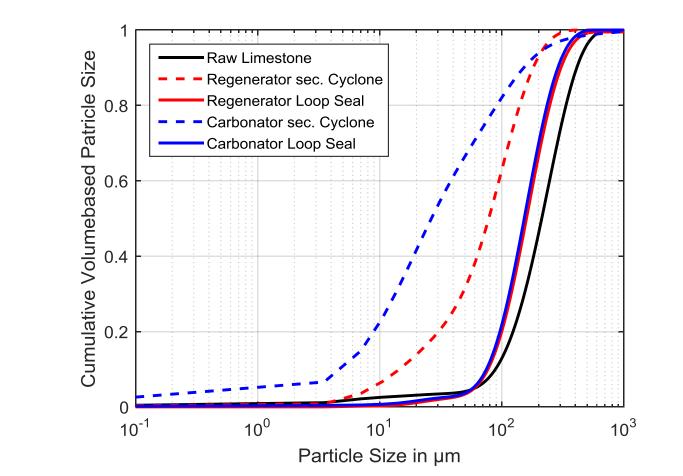
- Isothermal temperature profiles
- Stable hydrodynamics





Particle size

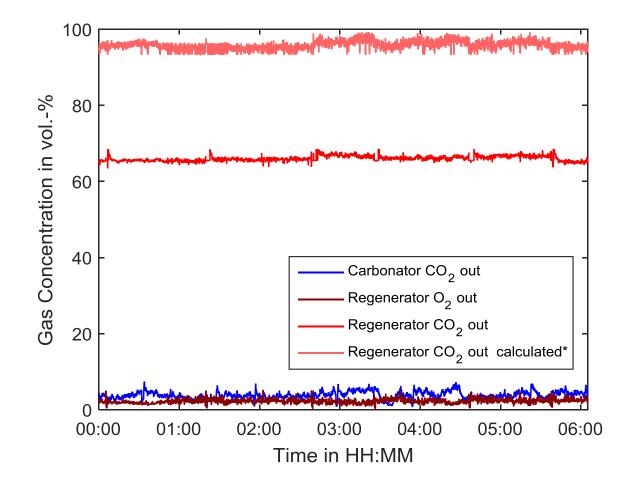
Reduced particle size due to attrition

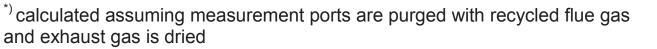


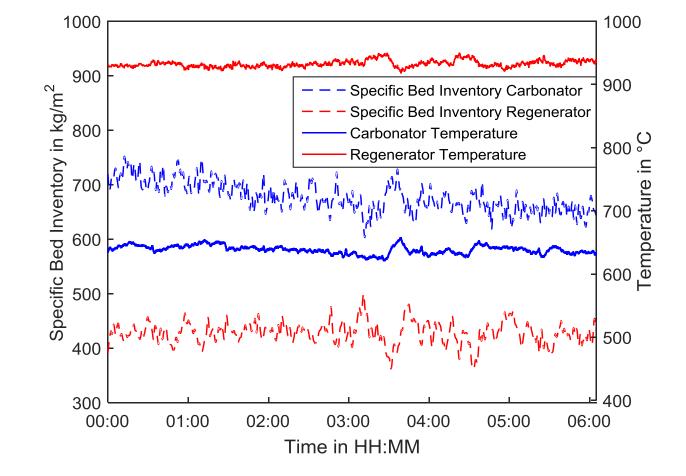
- Carbonation temperature: $T_{Carb} = 600 \dots 700 \ ^{\circ}C$
- Space time: $N_{Ca} / \dot{N}_{CO2} = 1.2 \dots 2 h$
- Makeup Ratio: $\dot{N}_{Ca,0} / \dot{N}_{CO2} = 0.5 ... 0.8$
- Looping Ratio: \dot{N}_{Ca} / \dot{N}_{CO2} = 7 ... 19
- At least 1 h of steady state operation for each experimental point
- Sorbent analysis by TGA as well as major and trace elementary analysis
- CO₂ flue gas concentration 15 % assuming cement plants raw meal is completely calcined in calcium looping

Process Demonstration

6 h extract of stable operation during 120 h experimental campaign







Temperature in °C Pressure in mbarg

Summary and Discussion

- Investigation of calcium looping process highly integrated into a cement plant resulting in CO₂ flue gas concentration of 15 % and high makeup ratios
- Stable operation was achived, yielding high CO₂ capture rates (> 90 % over wide parameter range)
- Uniform reaction condition in calciner and carbonator due to isothermal reactor temperature
- High CO₂ concentration in regenerator exhaust stream

Outlook

- Investigation of lower integrated calcium looping options resulting in
 - CO₂ concentrations up to 35 %
 - lower makeup ratios
- Measurement of impurities (N₂O, NO_x, etc.) in regenerator relevant for CO₂ compression



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Acknowledgements



