



The challenges with developing Primary Reference Materials for Hydrogen Purity

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MetroHyve/Hydraite Workshop

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Overview

- Purpose of Primary Reference Materials
- Why are there yet no PRMs for all impurities in Hydrogen
- Availability of PRMs
- Progress made in MetroHyve
- Remaining challenges

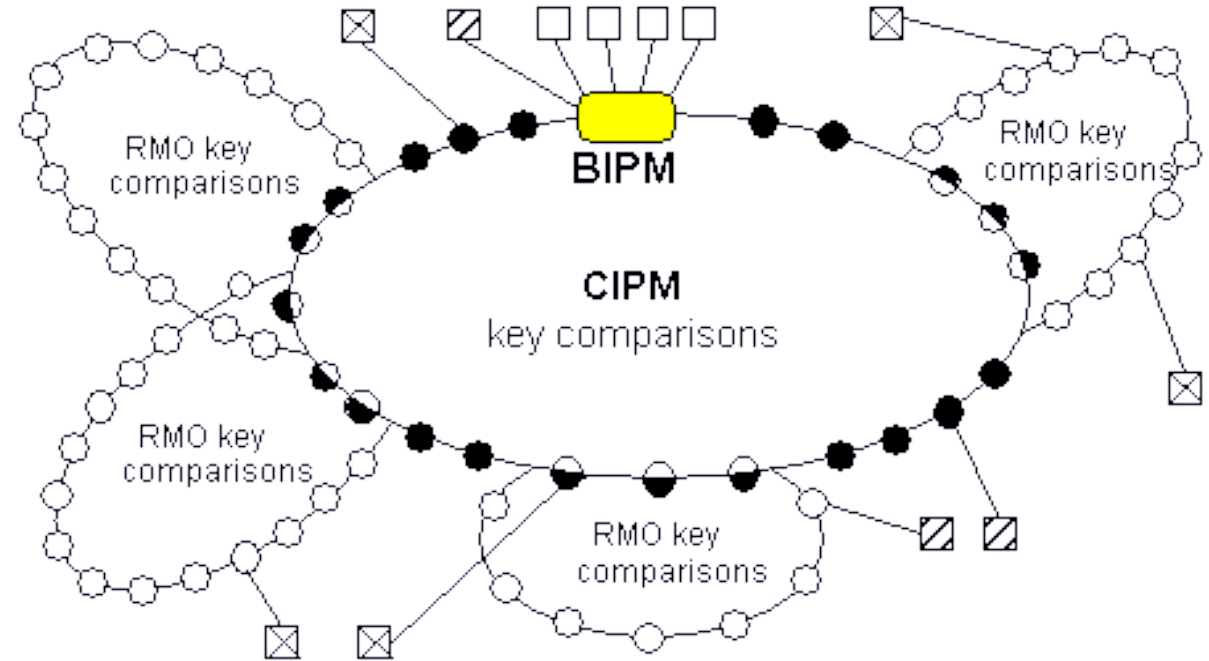
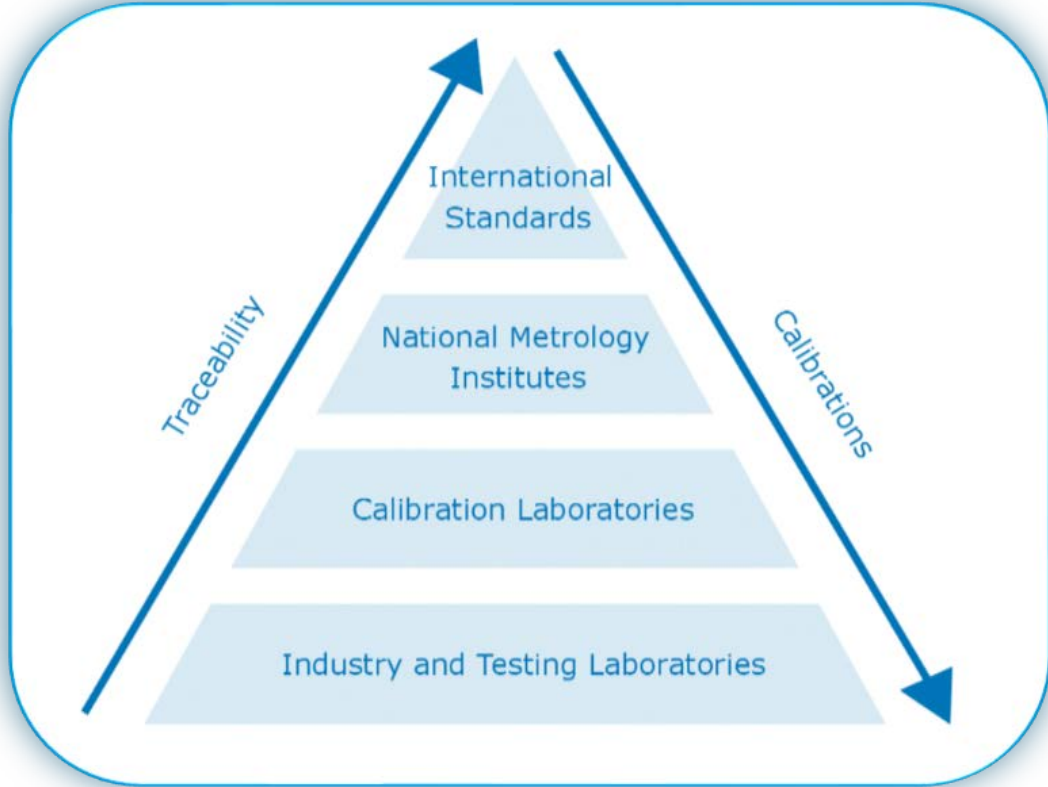


Measurement challenge

- Some of the impurities to be measured are at low concentration
- Some of the impurities are very reactive
- Classes of 'total concentration' like total halogenates are impossible for gravimetric preparation of gasmixtures

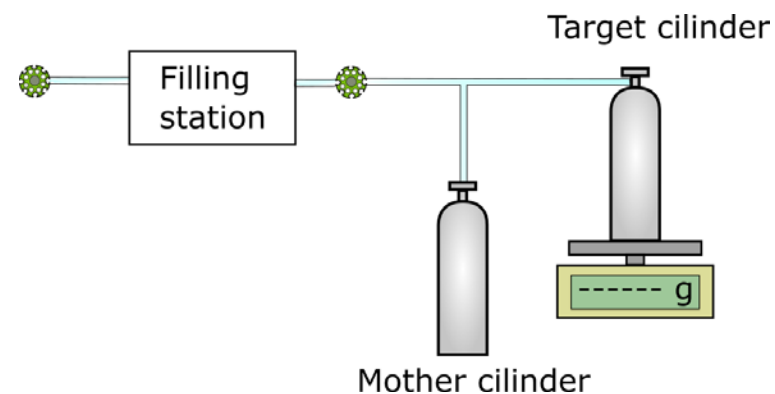
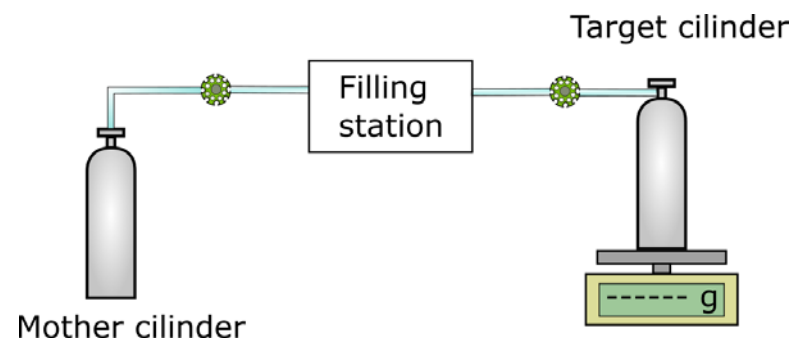


Metrological infrastructure



Preparation of Primary Reference Materials (1)

- Gravimetric preparation according to ISO6142
 - Thorough purity analysis of parent gases
 - Suitability of cylinders used
 - material compatibility, coatings, adsorption, stability
 - Verification of the prepared standards by accurate analytical methods



Preparation of Primary Reference Materials (2)

MetroHyve report A2.3.1

- Stability in Nitrogen does not mean stability in Hydrogen
 - 1 $\mu\text{mol/mol}$ CO in Nitrogen is stable in coated aluminium cylinders but 1 $\mu\text{mol/mol}$ CO in air has shown signs of instability
- Stability and concentration are closely linked
 - In Euramet 1220 a 1 $\mu\text{mol/mol}$ mixture of H_2S in H_2 showed 2 years stability, a test on 10 nmol of H_2S showed a large loss the day after preparation
- Even cylinders with the same treatment can behave very differently

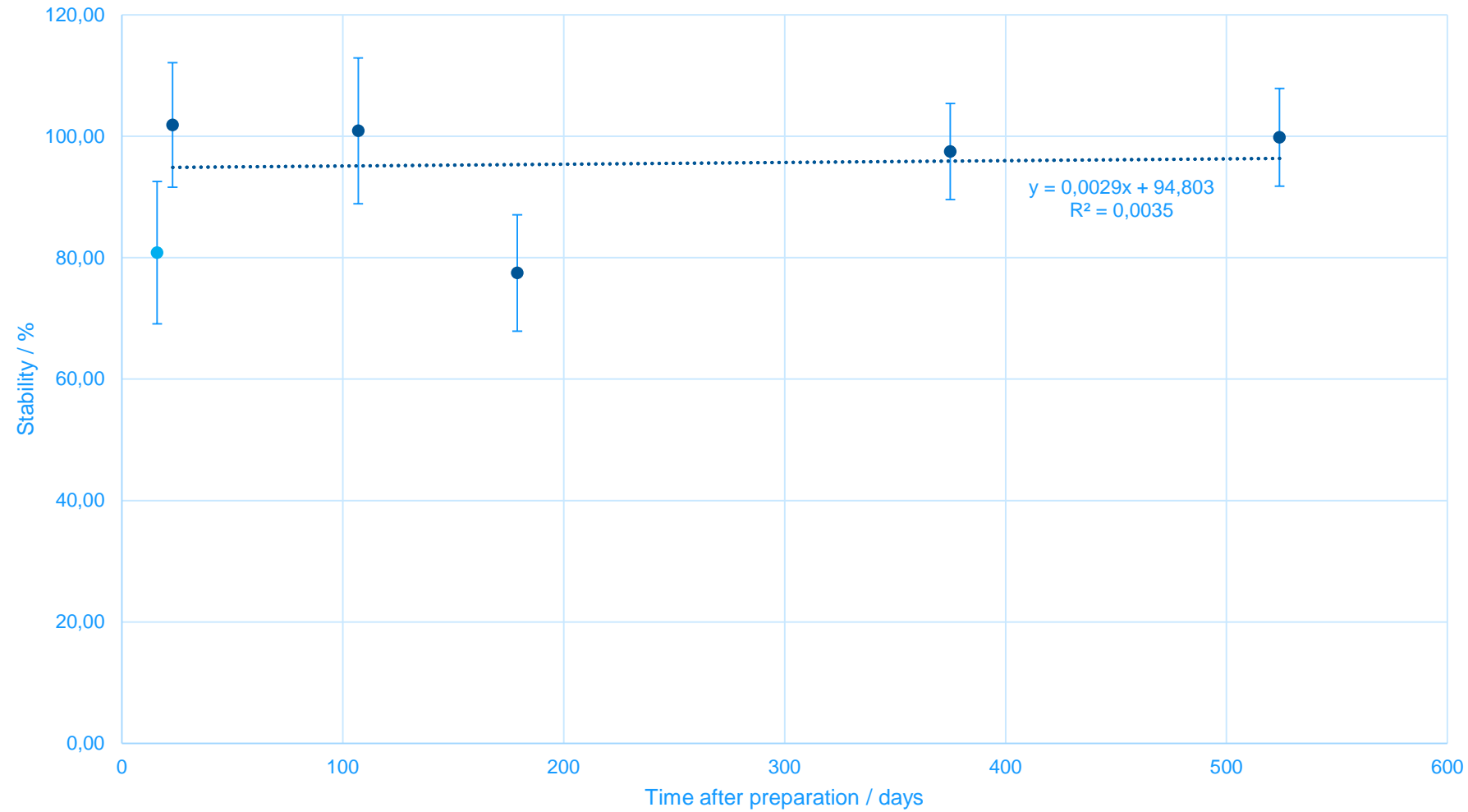
Accreditation: ISO17025 (calibration) ISO17034 (reference materials)

- Measurement capabilities in the BIPM KCDB
- Calibration entries by KRISS, NPL, VSL and VNIIM for hydrogen purity
- No entries for reference materials
- NPL has ISO17034 accreditation for stable components in Hydrogen

ISO 14687	
Water	Carbon monoxide
Total hydrocarbons	Total sulfur
Methane	Formaldehyde
Oxygen	Formic acid
Helium	Ammonia
Nitrogen	Halogenated compounds
Argon	
Carbon dioxide	

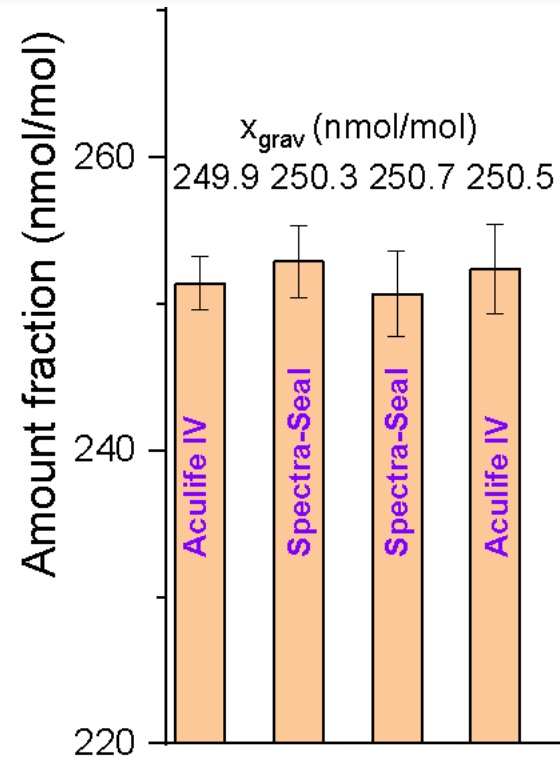
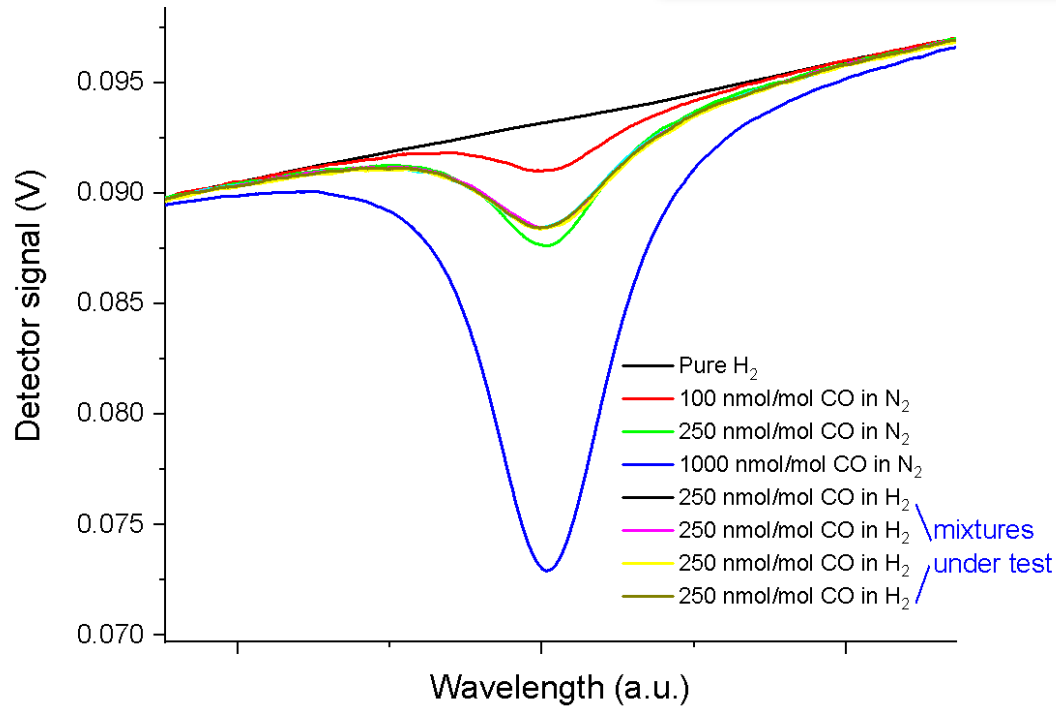


Stability for 40 ppb H₂S in H₂ mixtures at NPL

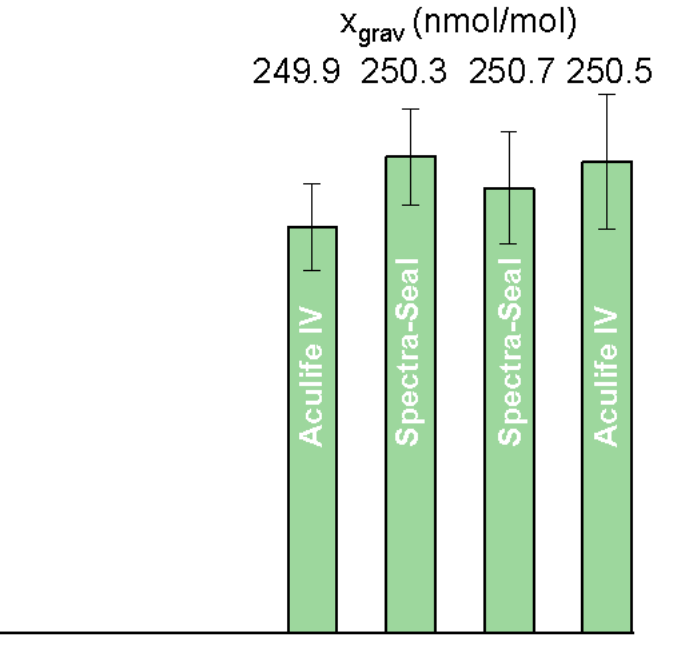


Short term stability for CO mixtures

CO in H₂ stability measurements



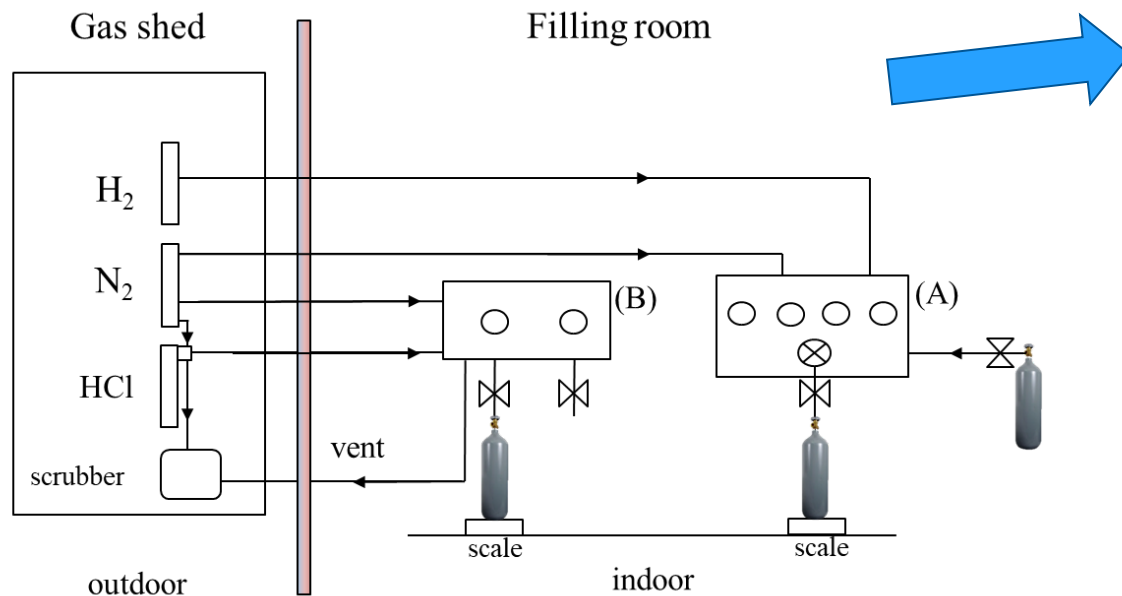
29 October
20 November



Mixture preparation dates: 7-16 August 2018

HCl in H₂ standards preparation by CEM

- By static gravimetric method (ISO 6142-1)
- 26 gas mixtures prepared in total
- Ranging from 10 µmol/mol to 4% mol/mol using dilution steps
- 5L Aluminium alloy cylinders



Key modifications of existing facilities at CEM

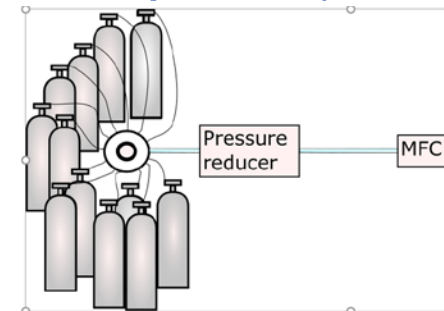
- HCl pure bottle placed outdoor ---- **safety**
- New filling station for HCl only, (B)
- New facilities allow purging with dry N₂ before HCl filling ---- **removing water inside lines**
- Vented HCl is conducted to a **water scrubber**

Analysis of the mixtures

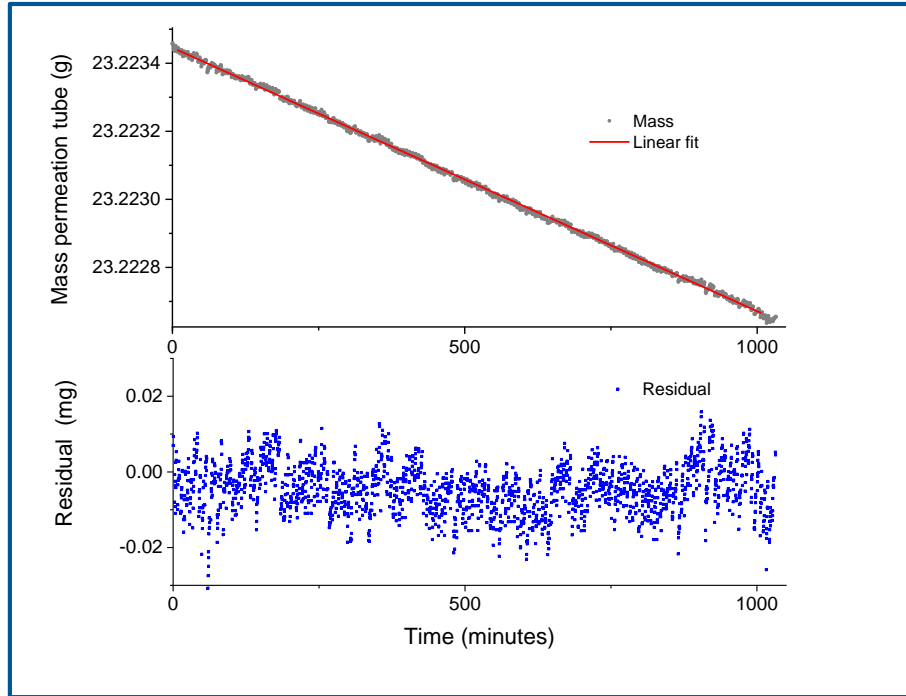
- Technique: Micro-GC-TCD
- Lowest concentration mixture analysed: 350 µmol/mol
- Still far away from ISO 14687-2 value: 0.05 µmol/mol

Short term stability test static HCl mixtures

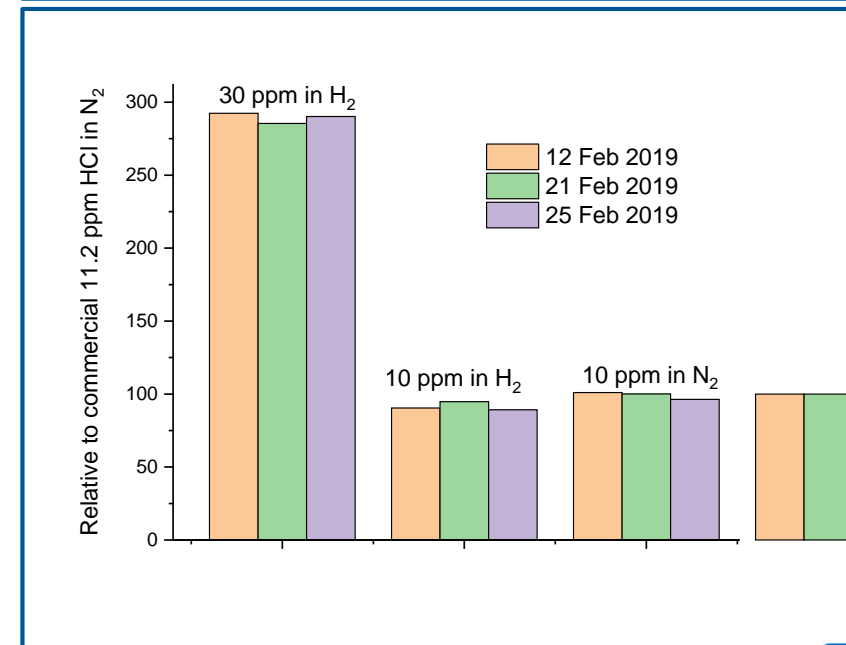
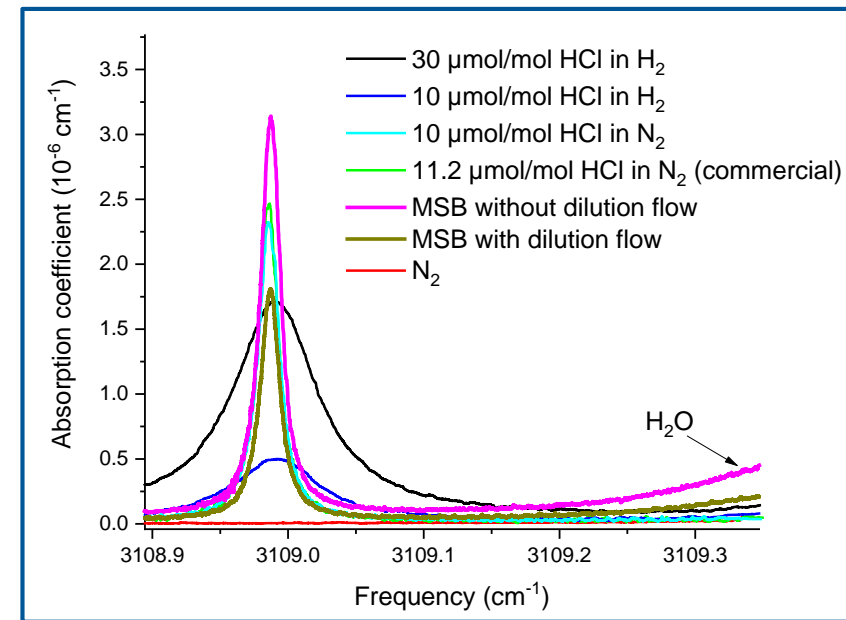
- Prepared in Aculife IV:
 - 10 and 30 $\mu\text{mol/mol}$ HCl in H_2
 - 10 $\mu\text{mol/mol}$ HCl in N_2
- Followed short term stability using CRDS (3 measurements in 9 days)
- Combined with MSB measurements
- Afterwards additional measurement using TDLS with spider (\Rightarrow HCl mixtures in H_2 now empty 😞)



Static gas mixtures HCl in H₂ at VSL



Analysis MSB data in progress to assign HCl amount fraction



Challenges

- Perform long term stability for CO and HCl in H₂ gas mixtures
- Development of PRMs for impurities like ammonia, formic acid and formaldehyde; the analytical methods developed are sensitive enough to be able to assess stability for those components
- What is the minimum shelf life acceptable to industry
- For NMIs: to perform comparisons to underpin the measurement infrastructure





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THANK YOU



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