Corrosion of pipeline materials due to impurities in separated CO₂ from fossil fuelled power plants

Institute of Energy Systems Benedikt Paschke Alfons Kather



Hamburg University of Technology Institute of Energy Systems Supported by:



Federal Ministry of Economics and Technology

6th Trondheim Conference on CO₂ Capture, Transport and Storage, TCCS-6

Trondheim, Norway. June 14-16, 2011

on the basis of a decision by the German Bundestag

Index



- 1. Motivation
- 2. CO₂-Purity
- 3. Experimental Setup
 - 1. Test Rig
 - 2. Experimental Conditions
- 4. Results and Discussions
- 5. Conclusions
- 6. Outlook

Motivation

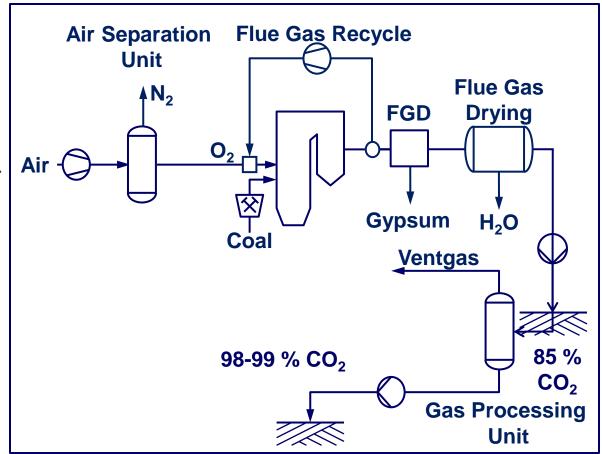


- Purity of separated CO₂ varies depending on the separation technology
 - ▶ Oxyfuel and Post-Combustion Capture \rightarrow oxidizing CO₂-compositions
 - ▶ Pre-Combustion Capture \rightarrow reducing CO₂-compositions
- Some impurities in the separated CO₂ can cause severe corrosion → purification may be required
- Higher purification correlates with higher expenditure of energy and/or capital
- Investigations are performed within the COORAL research project:
 - Overall objective: Define the required CO₂ purity for capture and storage (covering the whole chain from power plant to storage site)
 - Sub-objective: Determine the maximum permissible concentrations of impurities in the separated CO₂ stream for pipeline and compressor materials

CO₂-Purity



- Definition of different cases with varying gas compositions for experiments depending on the separation technology
- Three cases for Oxyfuel
 - ▶ "Zero Emission" (ZE)
 - "Purification"(Partial condensation)
 - "Rectification"
 (Partial condensation + Destillation)





Component	Zero Emission	Purification	Rectification
CO ₂	Rest	Rest	Rest
N ₂	5,8 Vol%	0,71 Vol%	100 ppmv
O ₂	4,7 Vol%	0,67 Vol%	100 ppmv
Ar	4,5 Vol%	0,59 Vol%	100 ppmv
H ₂ O	0-1000 ppmv	100 ppmv*	100 ppmv*
NO+NO ₂	100 ppmv	100 ppmv	100 ppmv
$SO_2 + SO_3$	70 ppmv	70 ppmv	70 ppmv
СО	50 ppmv	50 ppmv	50 ppmv

* Water content is limited to a maximum of 50–100 ppmv due to the low temperature of the purification process.

Experimental Setup



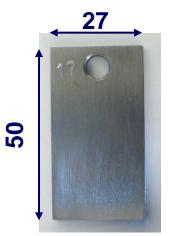
• Test rig

- 2 Hastelloy autoclaves (1I)
- 1 Autoclave (290 ml) with window
- I Syringe pump
- 1 Thermostat

• Experimental conditions

- Pressure: 110 bar
- ► Temperature: 60 °C
- Duration: 1 Week
- Materials
 - ► L360NB (1.0582)
 - ► L485MB (1.8977)

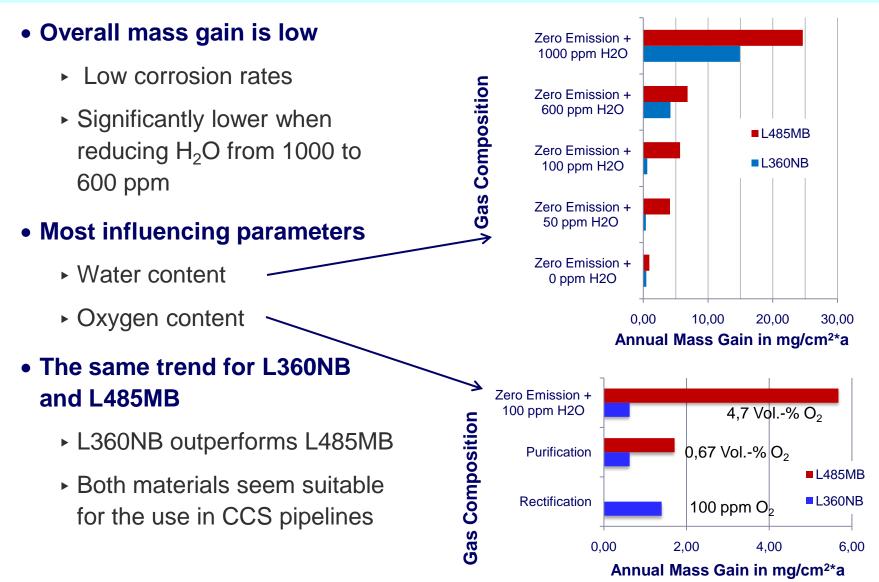






Results I





Results II



Oxide analysis with XRD

- Mixture of amorphous and crystalline species
- Only oxide being identified:
 α-FeOOH (Goethite)

Oxide analysis with SEM-EDX

- Primarily: Fe, O and little S
- Traces of: C, N
- Only uniform corrosion detected
 - No signs of pitting

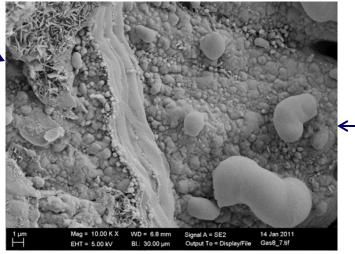
0 ppm H₂O 10

$100 \text{ ppm } \text{H}_2\text{O} \quad 1000 \text{ ppm } \text{H}_2\text{O}$





L485MB specimens in "Zero Emission"composition with varying water content after 1 week

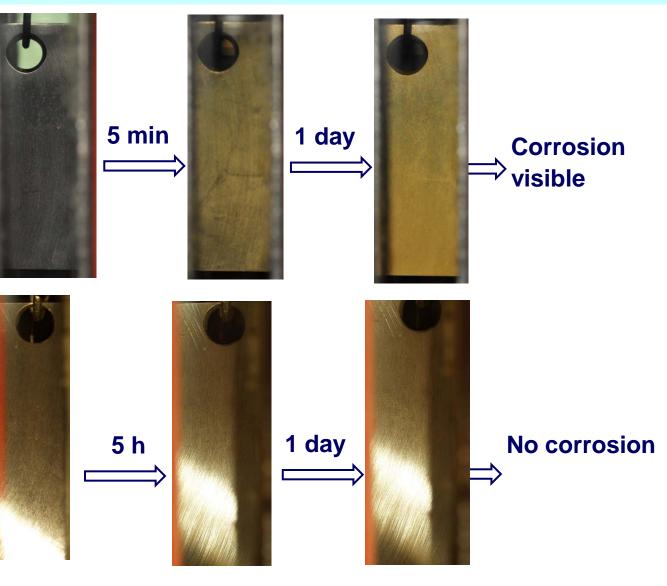


SEM Image of the surface of a corroded L485MB specimen

Influence of different Impurities



• Zero Emission + 600 ppm H₂O



CO₂ + 1000 ppm H₂O + 4,7 Vol.-% O₂

6th Trondheim Conference on CO₂ Capture, Transport and Storage, TCCS-6 Trondheim, Norway. June 14-16, 2011

Conclusions



- Either NO or SO₂ was required to initiate corrosion
- Traces of CO have only little or no effect on corrosion
- At least one of the following components should be minimized to reduce corrosion: H₂O, O₂ or acid gas components (SO_x and NO_x)
- Water content should not exceed 600 ppm, 100 ppm or lower is beneficial
- L360NB and L485MB seem to be suitable for transporting impure CO₂ from Oxyfuel processes, whereas L360NB is advantageous

Outlook



- Investigation of further materials (including compressor materials)
- Addition of post and pre combustion capture cases
 - Corrosion is expected to be lower because impurities are less for both technologies
- Determination of the influence of fluid flow
- Analysis of oxides and corrosion mechanism



Thank you for your attention!

Contact: Benedikt.paschke@tuhh.de

6th Trondheim Conference on CO_2 Capture, Transport and Storage, TCCS-6 Trondheim, Norway. June 14-16, 2011