

Combining CO₂ streams from different emitters - a challenge for pipeline transportation



Le Quynh Hoa¹, Sandra Knauer², Dirk Bettge¹, Ralph Bäßler¹, Axel Kranzmann¹, Heike Rütters³

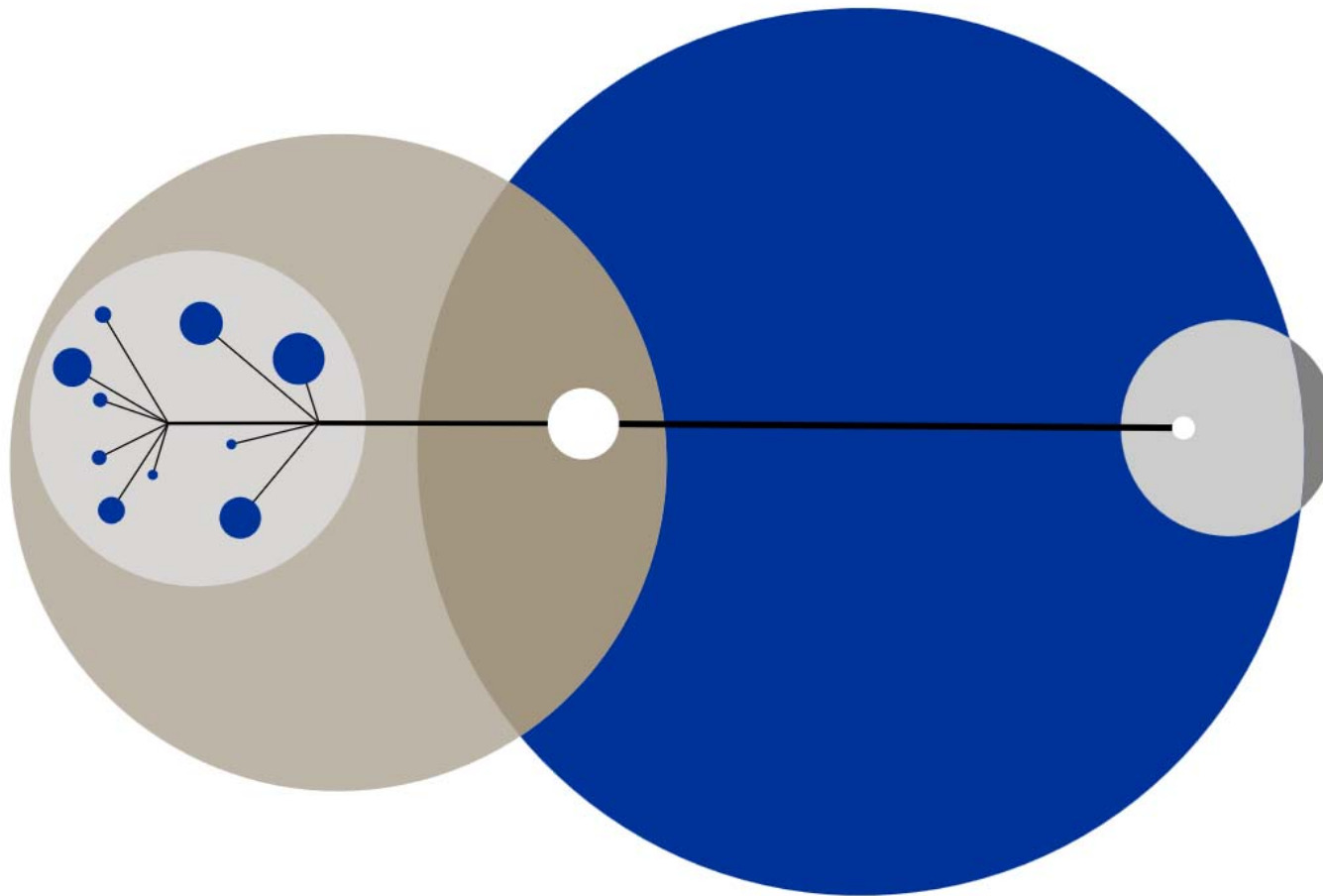
¹ BAM Federal Institute for Materials Research and Testing, Unter den Eichen 87, D-12205 Berlin, Germany

² Eurotechnica GmbH, An den Stücken 55, D-22941 Bargteheide, Germany

³ Federal Institute for Geosciences and Natural Resources (BGR), Stilleweg 2, D-30655 Hannover, Germany

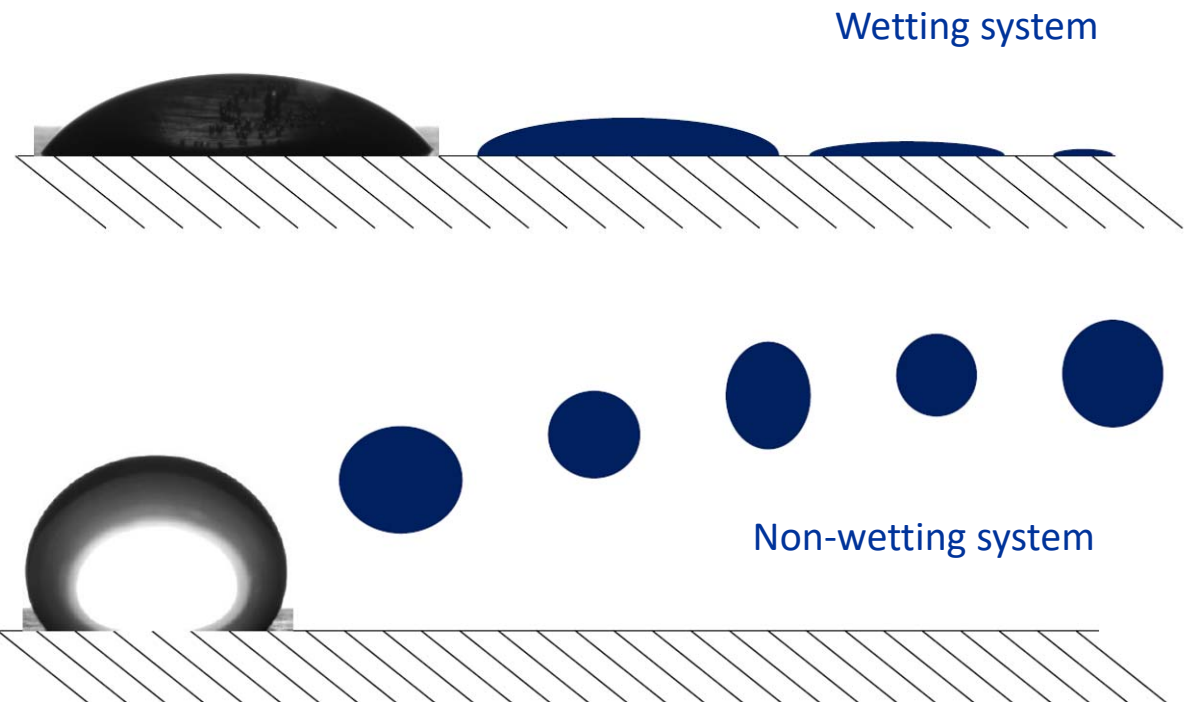
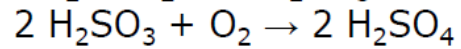
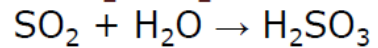
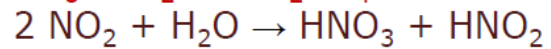
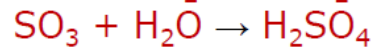
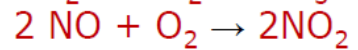
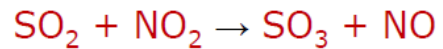


Cluster Scenario

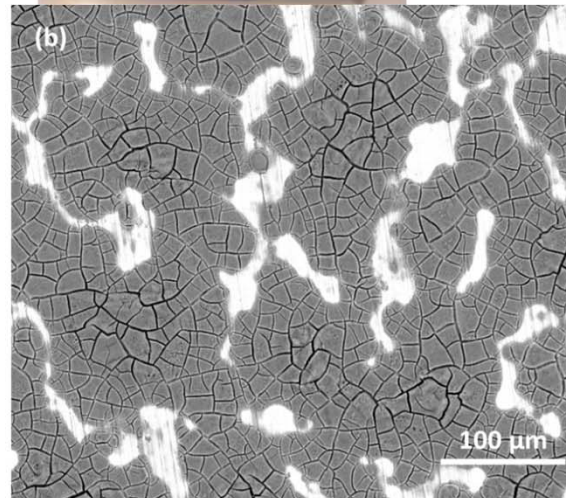
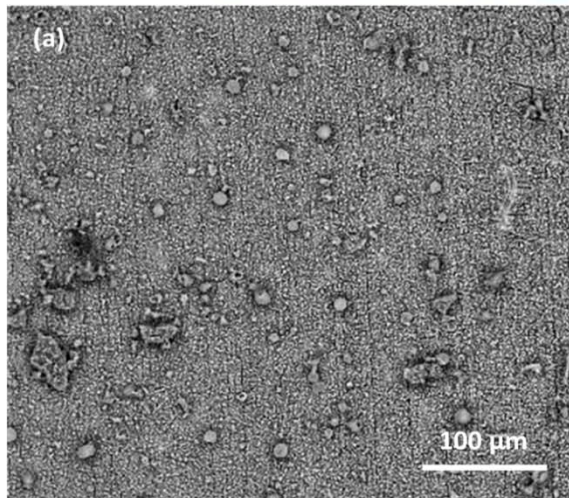
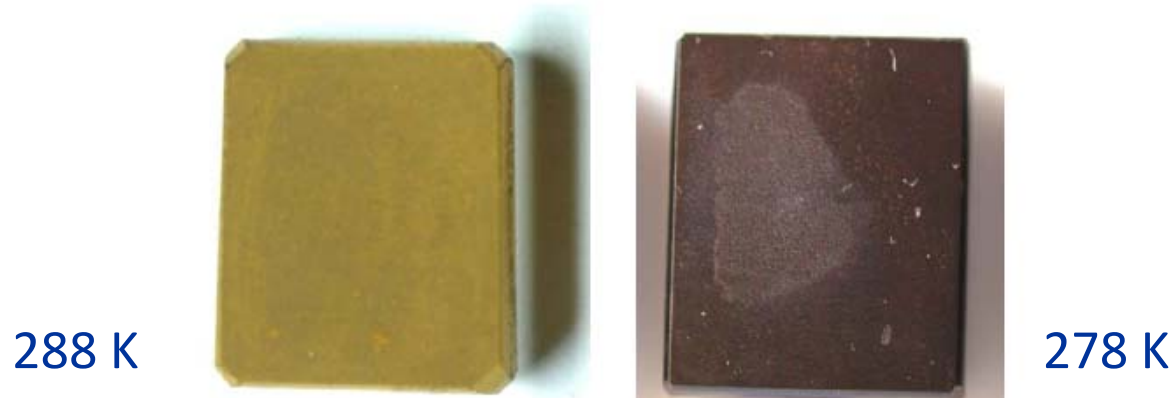


Not to scale

Reactions in CO₂ streams – acid formation and condensation



Condensation of water and/or acids: temperature effect



Backscattered electron imaging

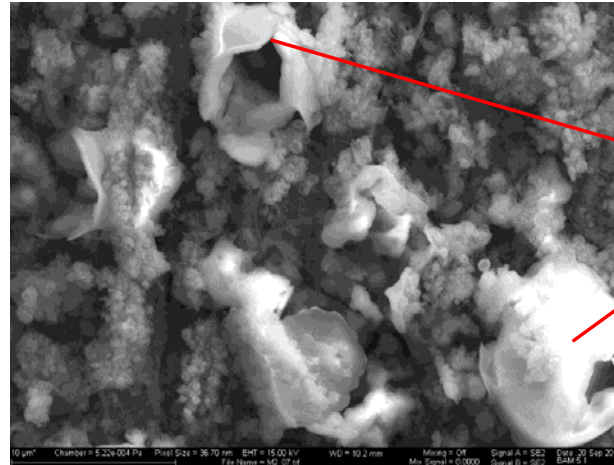
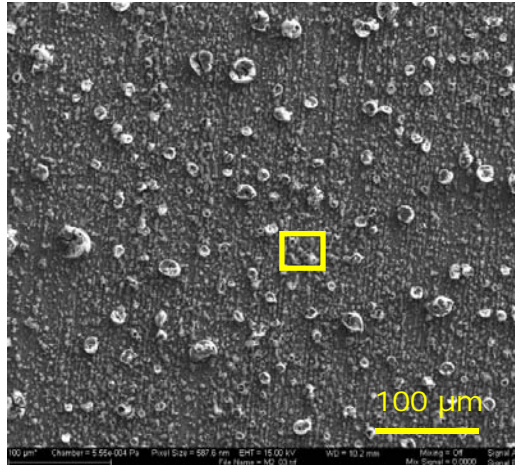
X70 (L485MB)

in CO₂ with
200 ppm_{mole} H₂O,
220 ppm_v SO₂,
6700 ppm_v O₂

@ 5 MPa, 7 days

Cross-reactions – implications for corrosion (X70)

SEM imaging



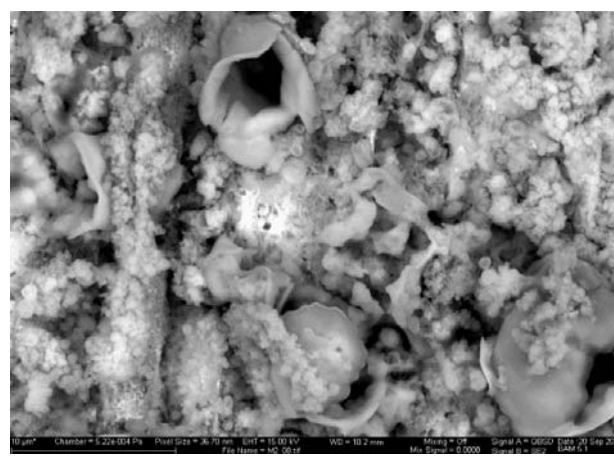
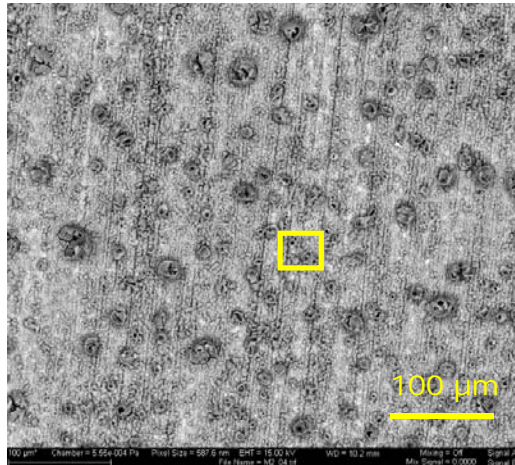
Ratio of O/S is higher cf. product formed without NO_2

X70 (L485MB)

in CO_2 with
200 $\text{ppm}_{\text{mole}} \text{H}_2\text{O}$,
70 $\text{ppm}_v \text{SO}_2$,
100 $\text{ppm}_v \text{NO}_2$
and 6700 $\text{ppm}_v \text{O}_2$

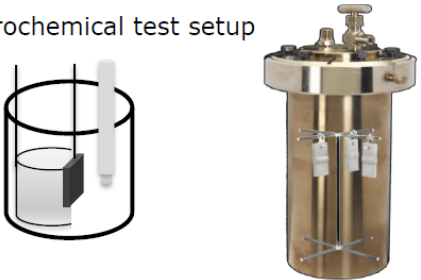
after 7 days
@ 5 MPa, 278 K

Backscattered electron imaging



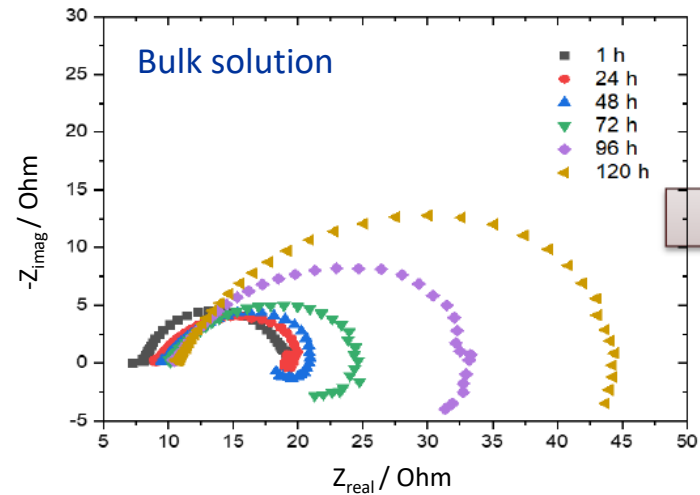
Approaches to study acid corrosion

Electrochemical test setup



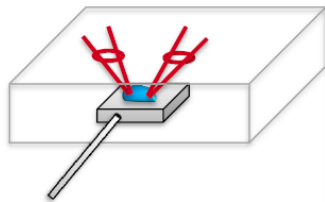
500 mL CO₂-saturated Synthetic Condensate

Traditional Approach

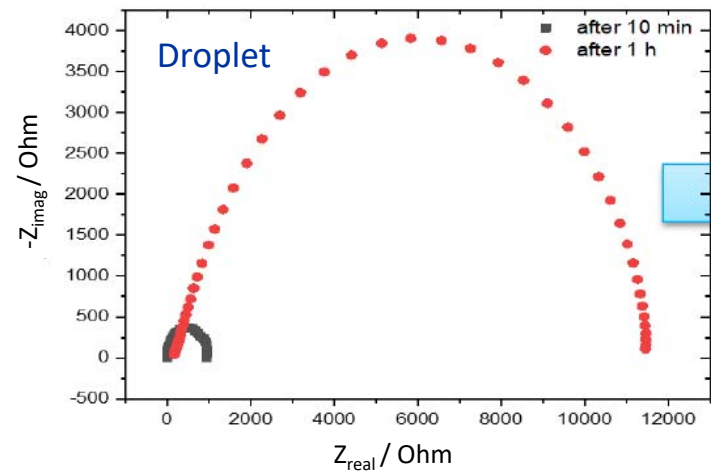


- dissolvable corrosion products
- high corrosion rate
- strong pitting

Counter- and reference electrodes

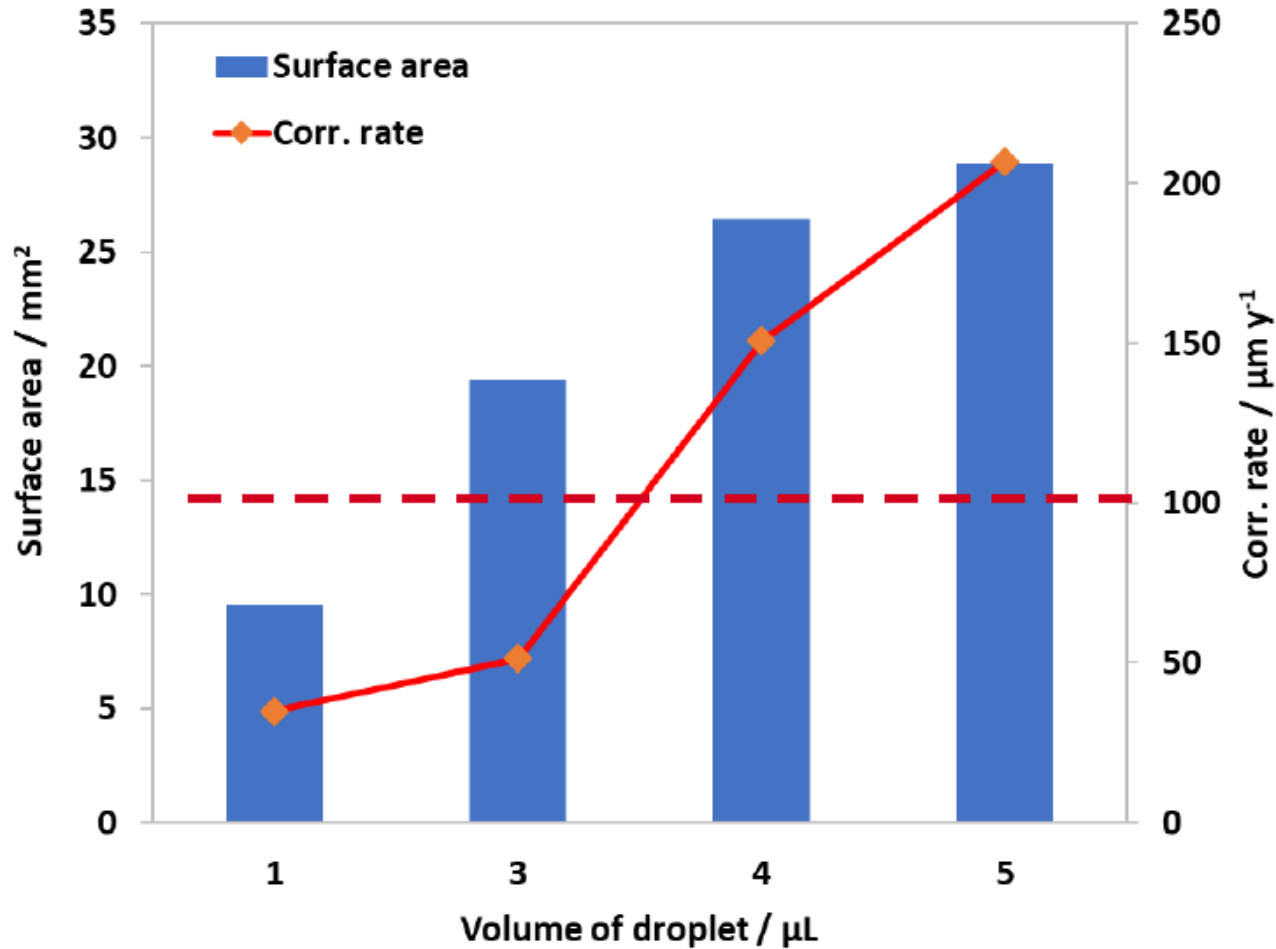


A drop on metal surface in CO₂



- thick, high-resistance corrosion products
- low corrosion rate

Effect of droplet size

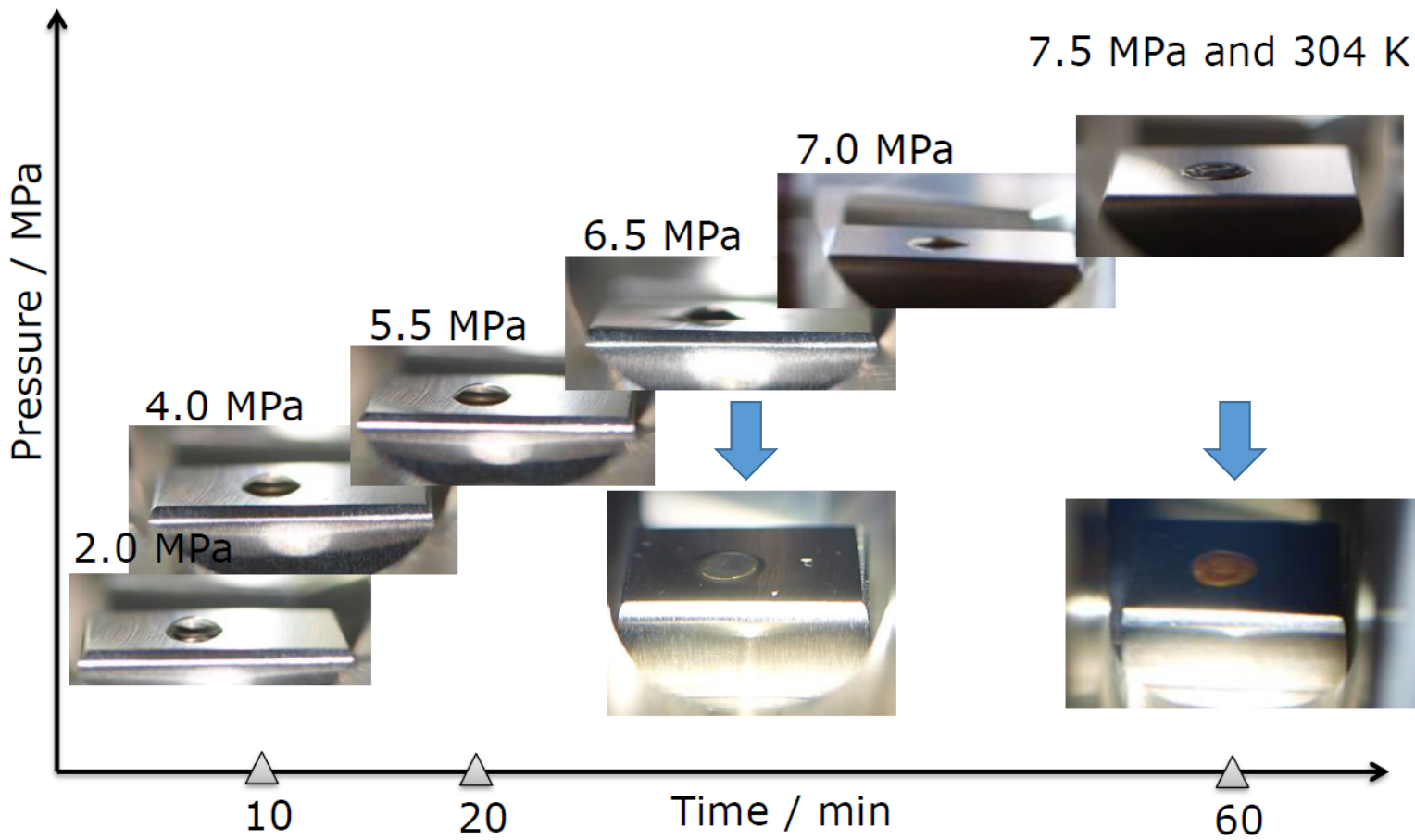


X70 (L485MB)

in CO_2 with
220 ppm_v SO_2 and
6700 ppm_v O_2

@ 278 K, 5 MPa,
7 days

Acid formation by impurity diffusion into CO₂-saturated water drop



X70 (L485MB)

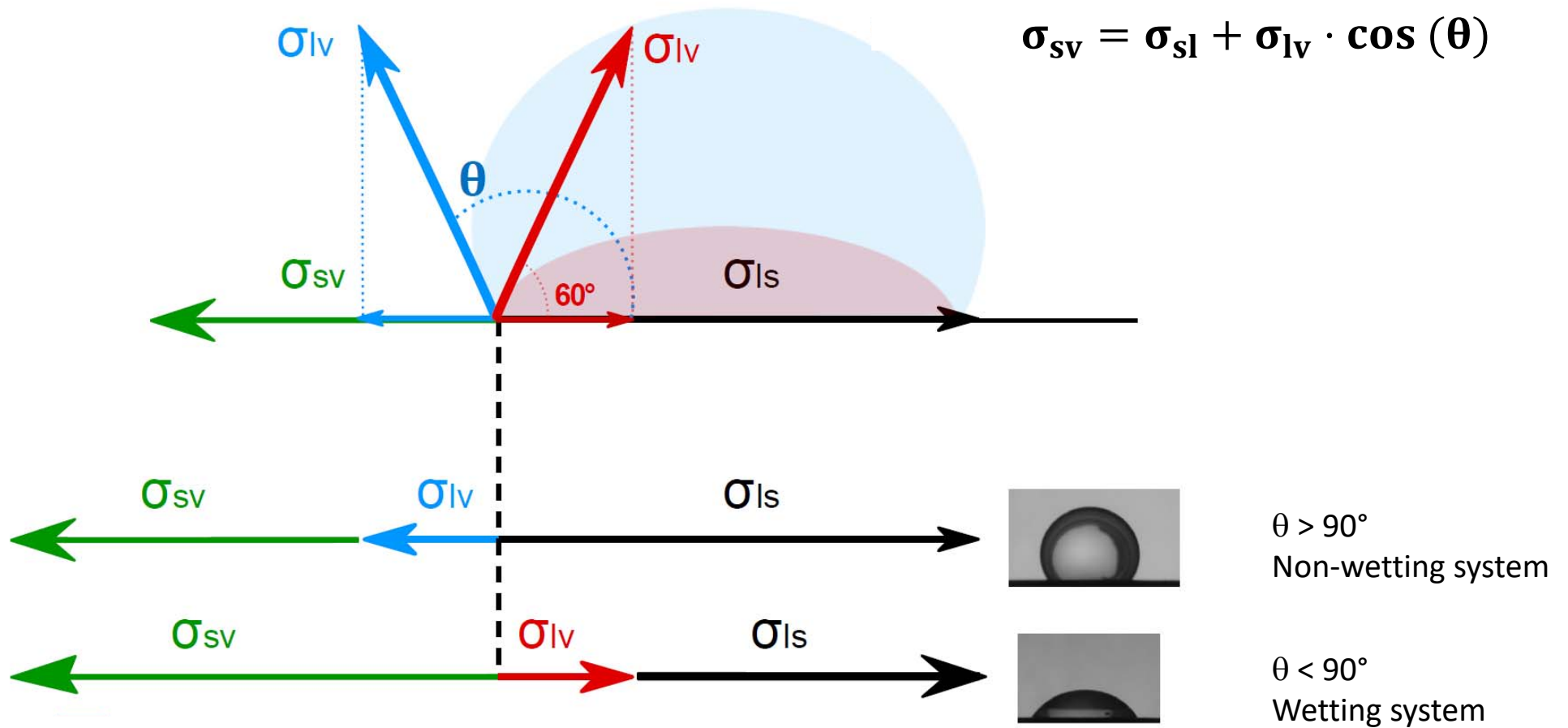
in CO₂ with
220 ppm_v SO₂ and
6700 ppm_v O₂

V_{drop} = 5 μL

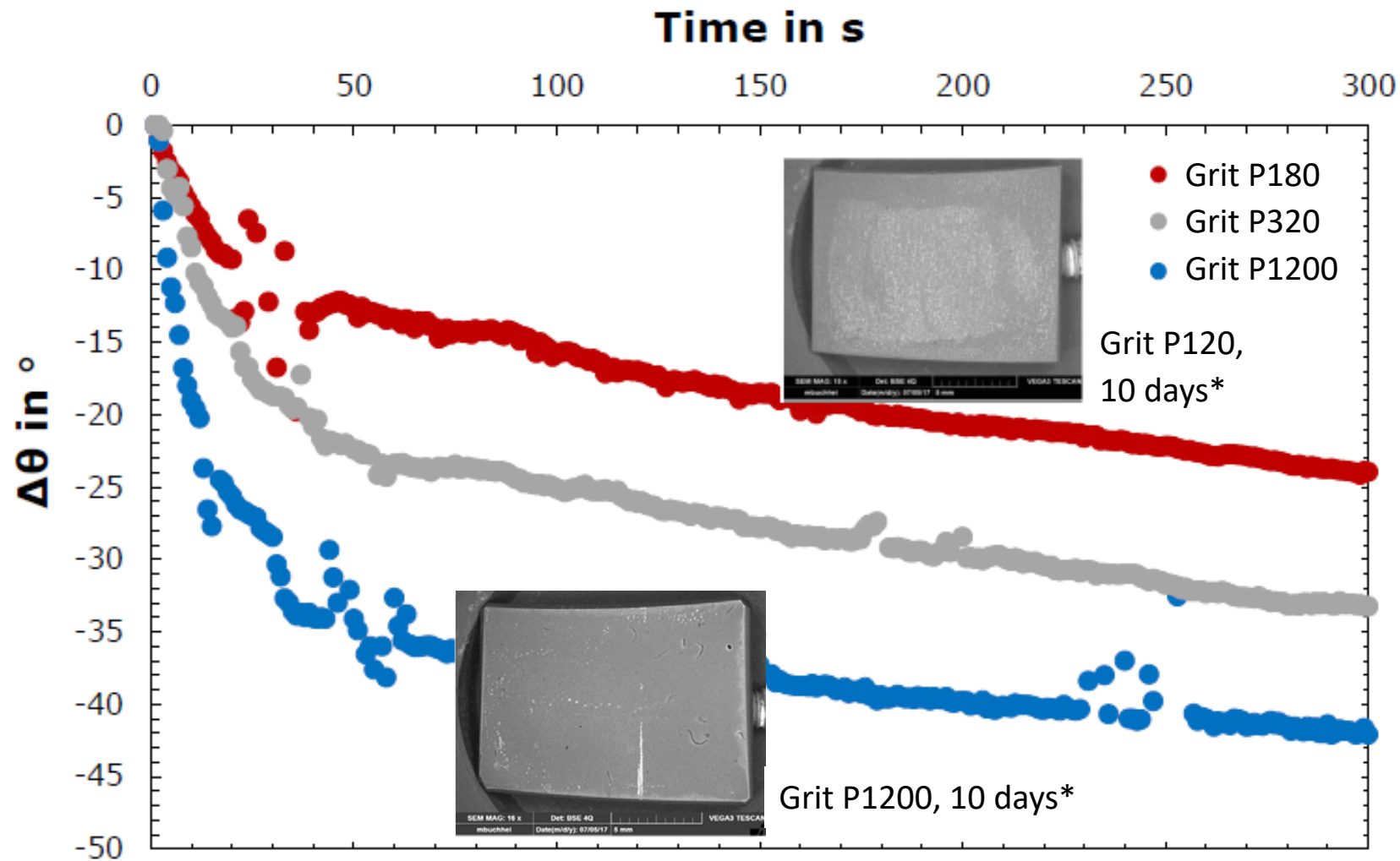
Wettability: Contact angle

Young equation:

$$\sigma_{sv} = \sigma_{sl} + \sigma_{lv} \cdot \cos(\theta)$$



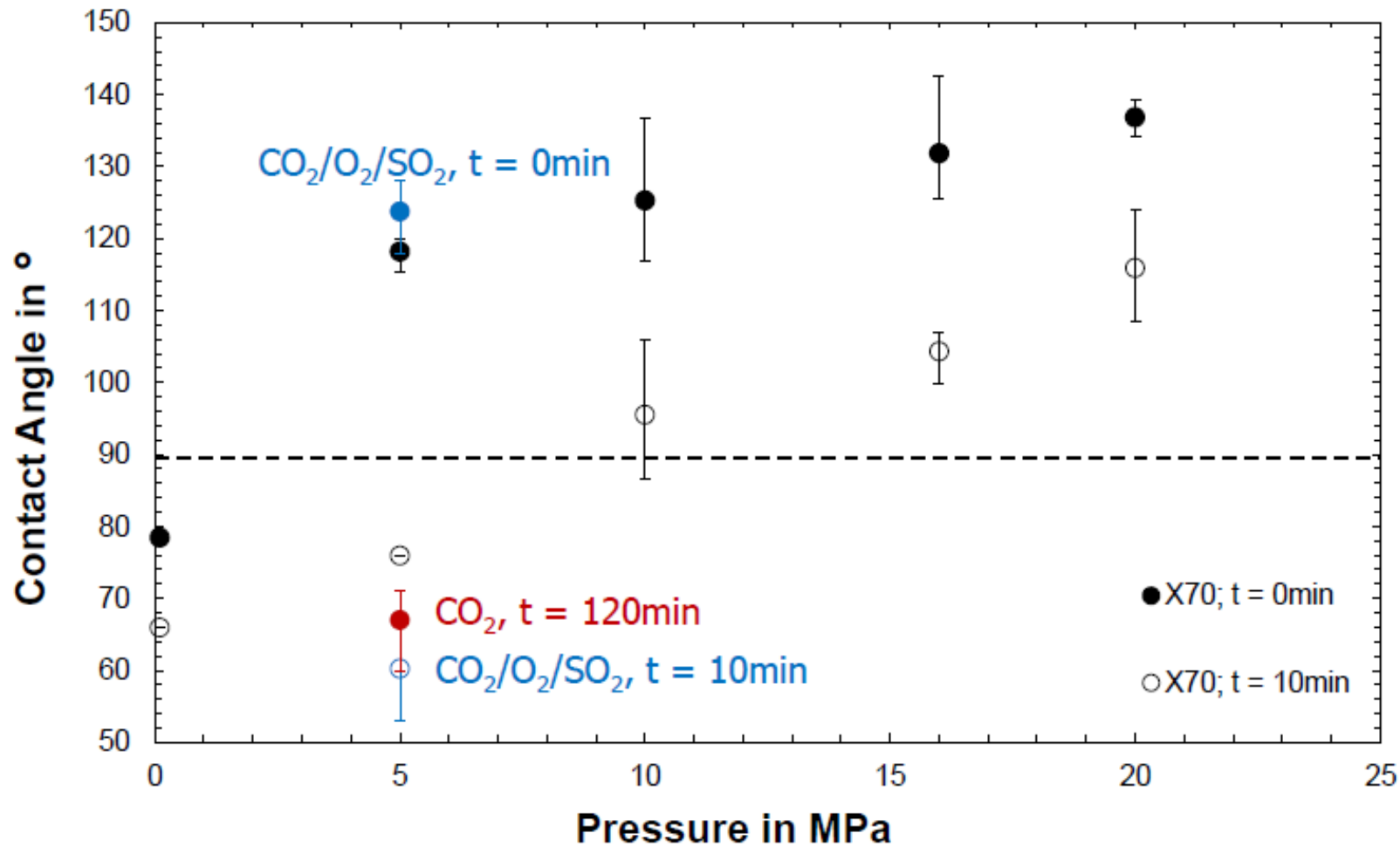
Contact angle difference ($\Delta\theta$) as f (surface roughness)



X70 (L485MB)
in CO₂
@ 278 K,
5 MPa

*: coupons – exposed to CO₂ for 10 days at 278 K, 5 MPa with SO₂ (220 ppm_v), O₂ (6700 ppm_v), H₂O (200 ppm_{mole})

Contact Angle as f (p, impurities)



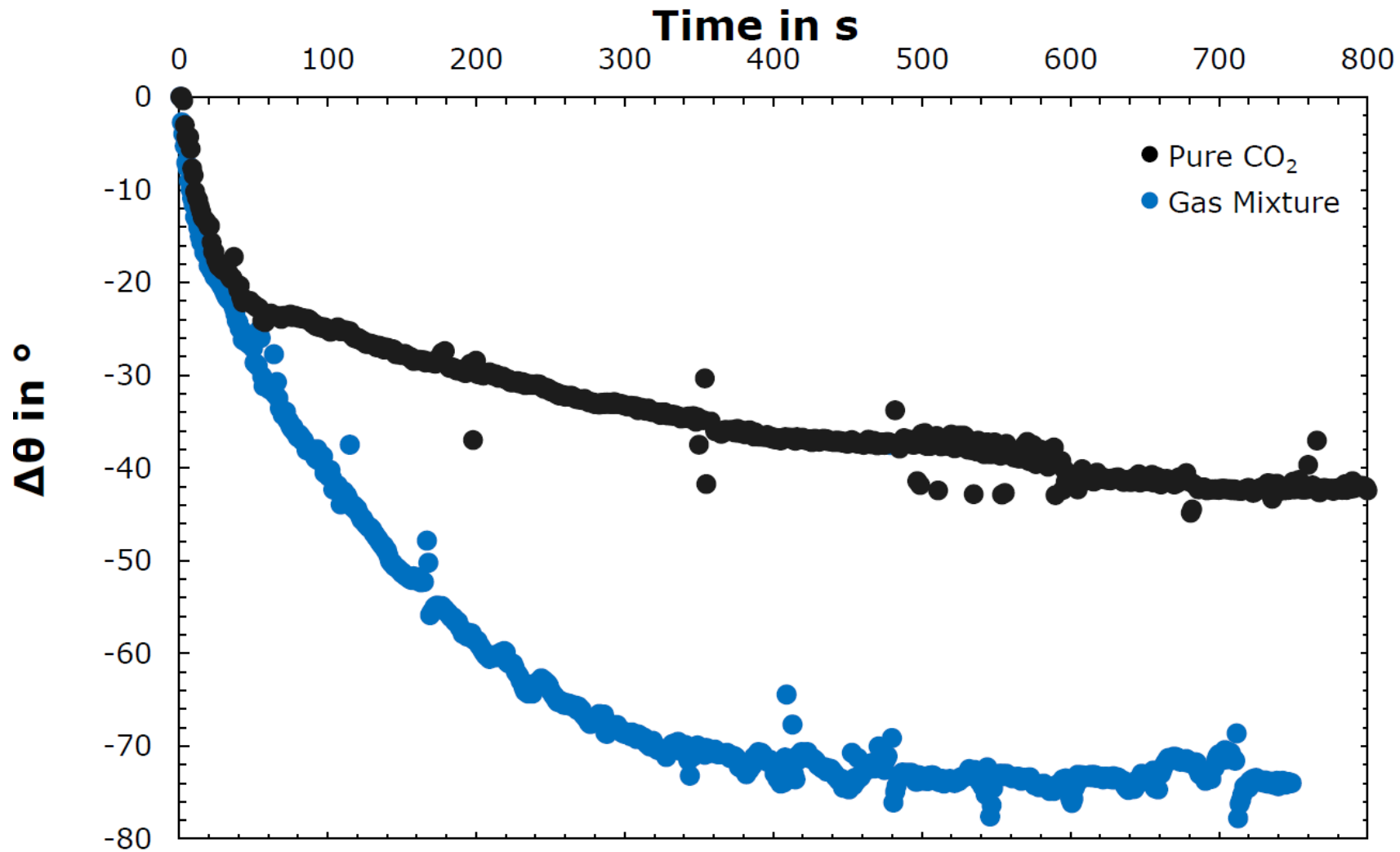
X70 (L485MB)

in CO₂
with
220 ppm_v SO₂ /
6700 ppm_v O₂

@278 K

(Grit P320)

Contact angle changes – reactive wetting (X70)



X70 (L485MB)
in CO₂ (with
220 ppm_v SO₂,
6700 ppm_v O₂)

@ 278 K, 5 MPa

(Grit P320)

In summary

Contact angle decreases

....with decreasing pressure,
....with time due to formation of corrosion products,
....faster on smoother surfaces,
....faster in case of higher corrosion rate
(e.g. in presence of O_2 , SO_2)

resulting in better wetting and thus potentially in ...

.....larger contact area droplet/steel,
.....droplet mobilization being less likely,
.... droplet accumulation being more likely,
..... potentially more severe corrosion.





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