

A Heat Integrated Solid-sorbent Based Fluidized Bed Process for Post-Combustion CO₂ Capture

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Introduction

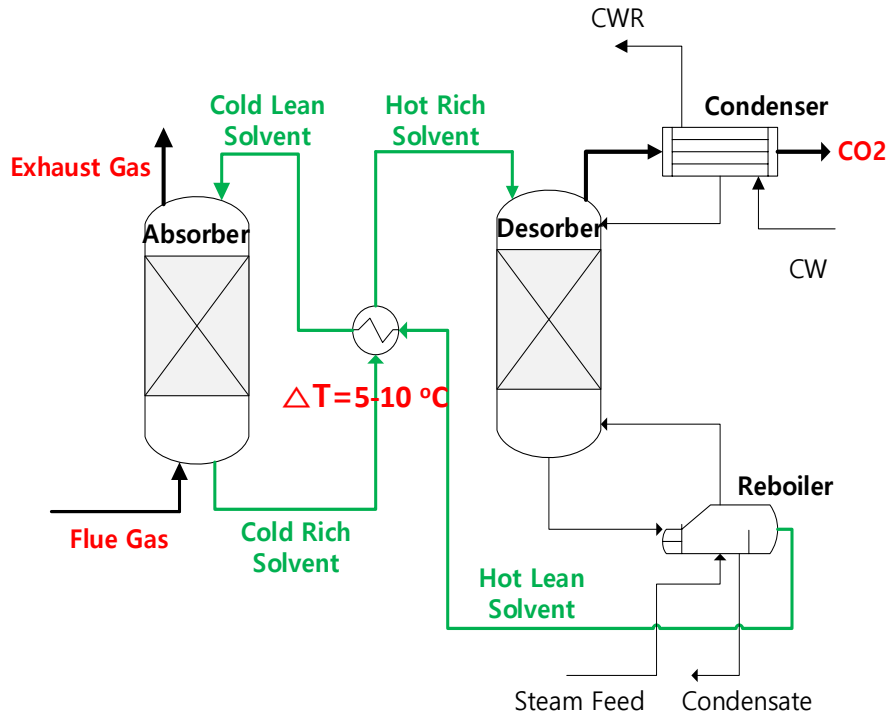
Process description

Features of developed solid sorbent

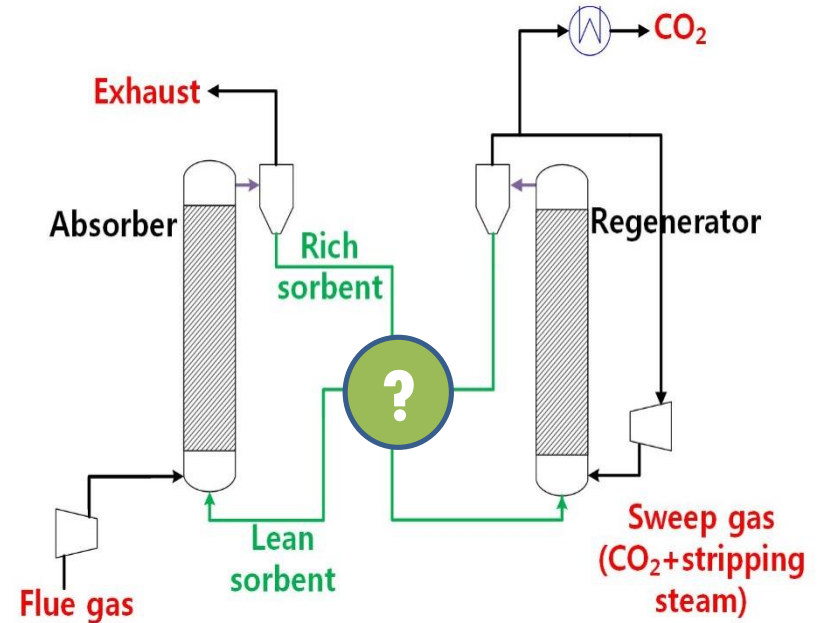
Process demonstration

Summary

Amine Scrubbing Process

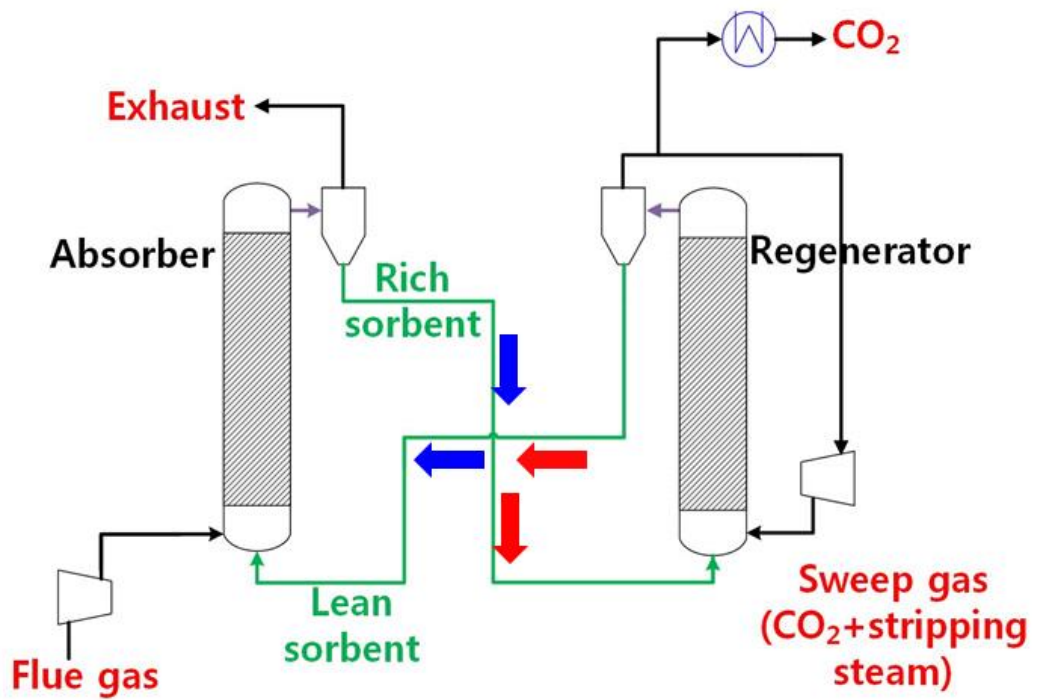


Solid Sorbent-based Process

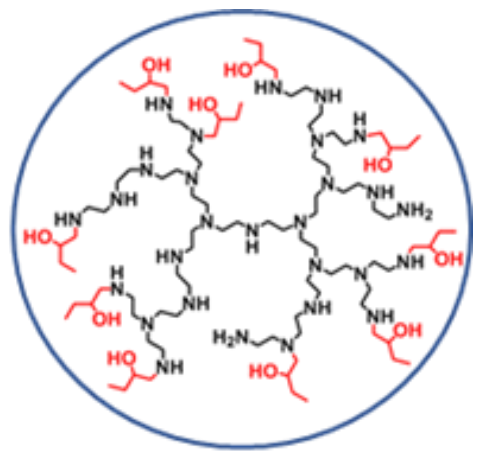


- **Advantage of dry-sorbent process**
 - Low degradation / corrosiveness / volatility
- **Main issue**
 - 1) Difficult heat exchange between lean/rich sorbents
 - 2) Low fluidity

- Low temperature process with sensible heat exchange



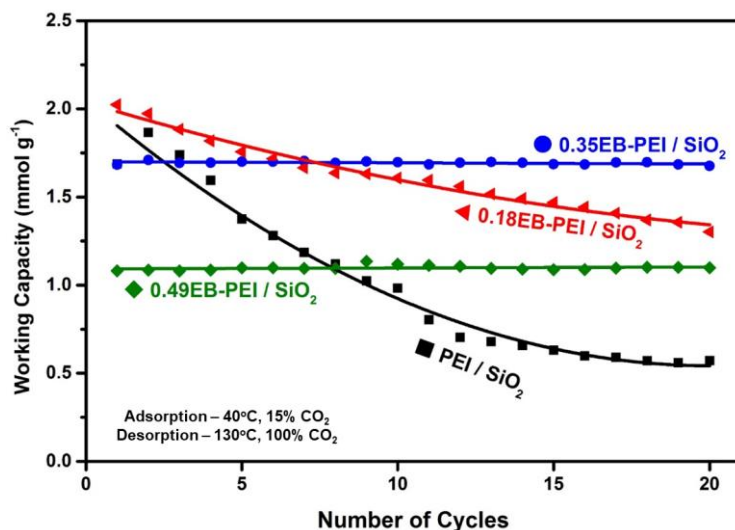
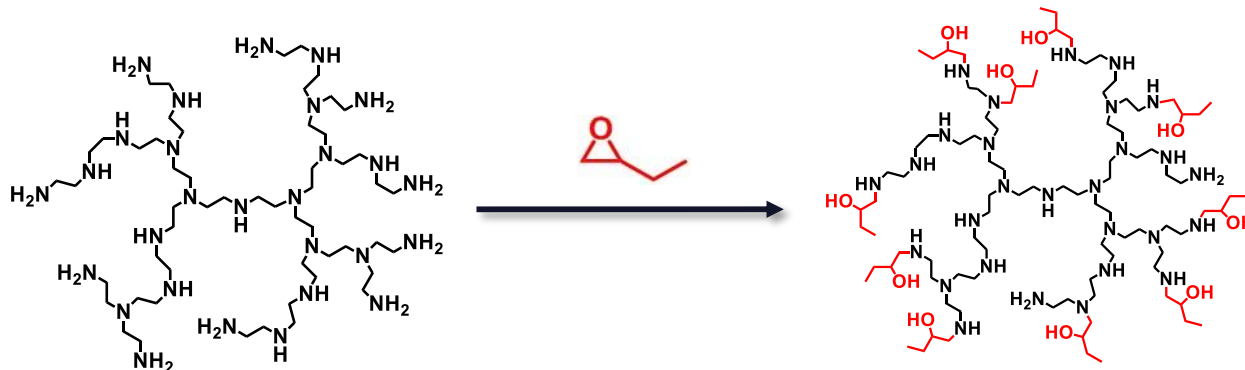
- Epoxy functionalized sorbent with a low heat of absorption



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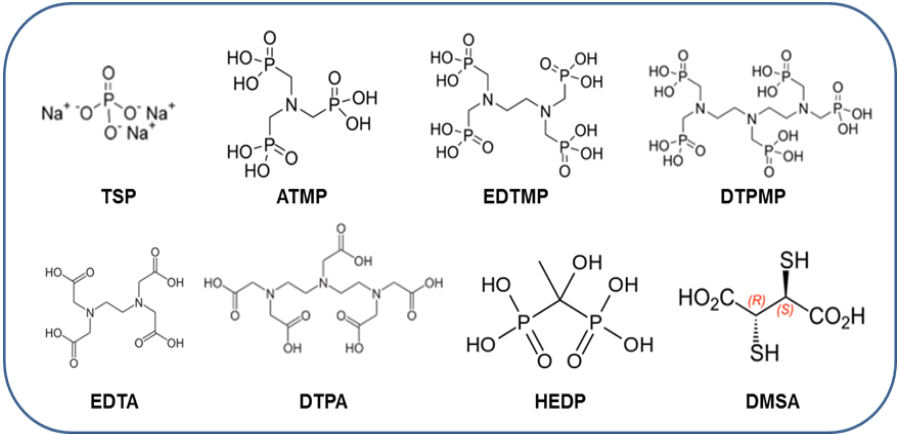
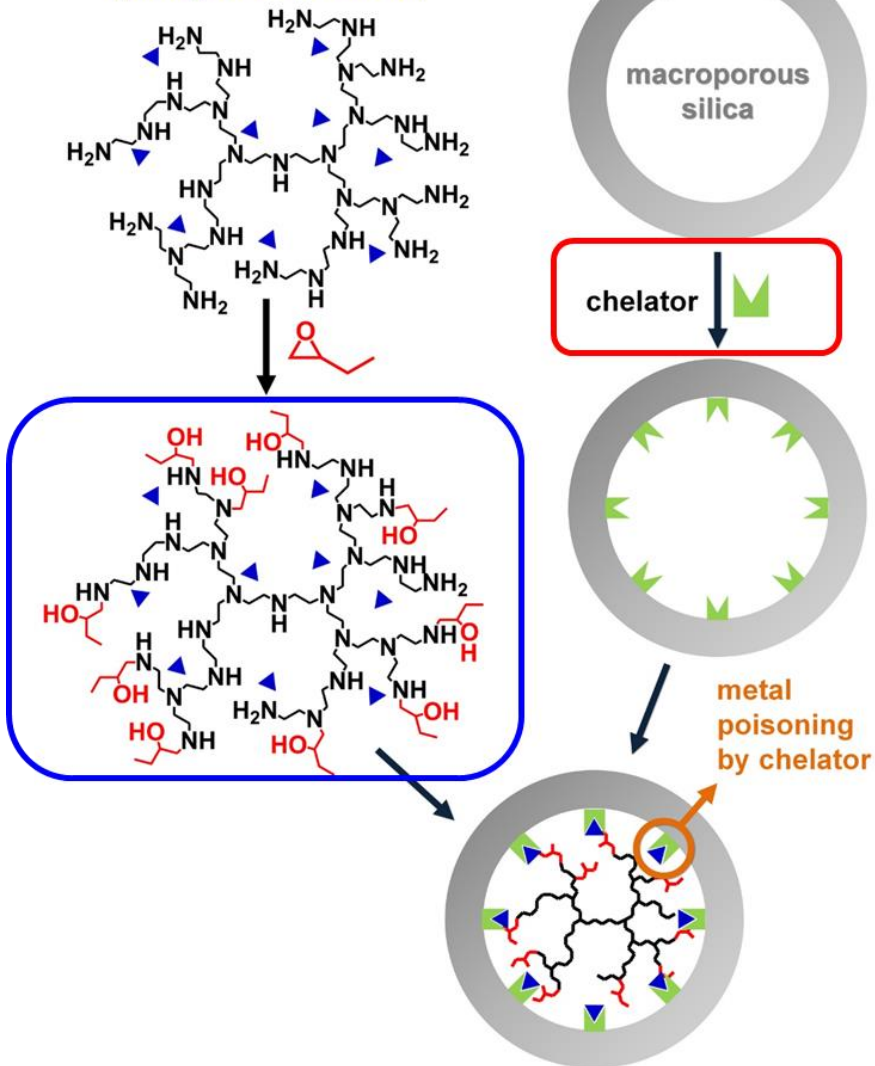
<p>KRICT (Dr. Yong Ki Park)</p> <ul style="list-style-type: none">- Process development- Sorbent shaping & scale-up	<p>KAIST (Prof. Minkee Choi)</p> <ul style="list-style-type: none">- Sorbent development
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- Epoxy-functionalization of primary amines in PEI
 - Resistant to urea formation (less steam demand in a desorption bed)
 - Low heat of CO₂ absorption



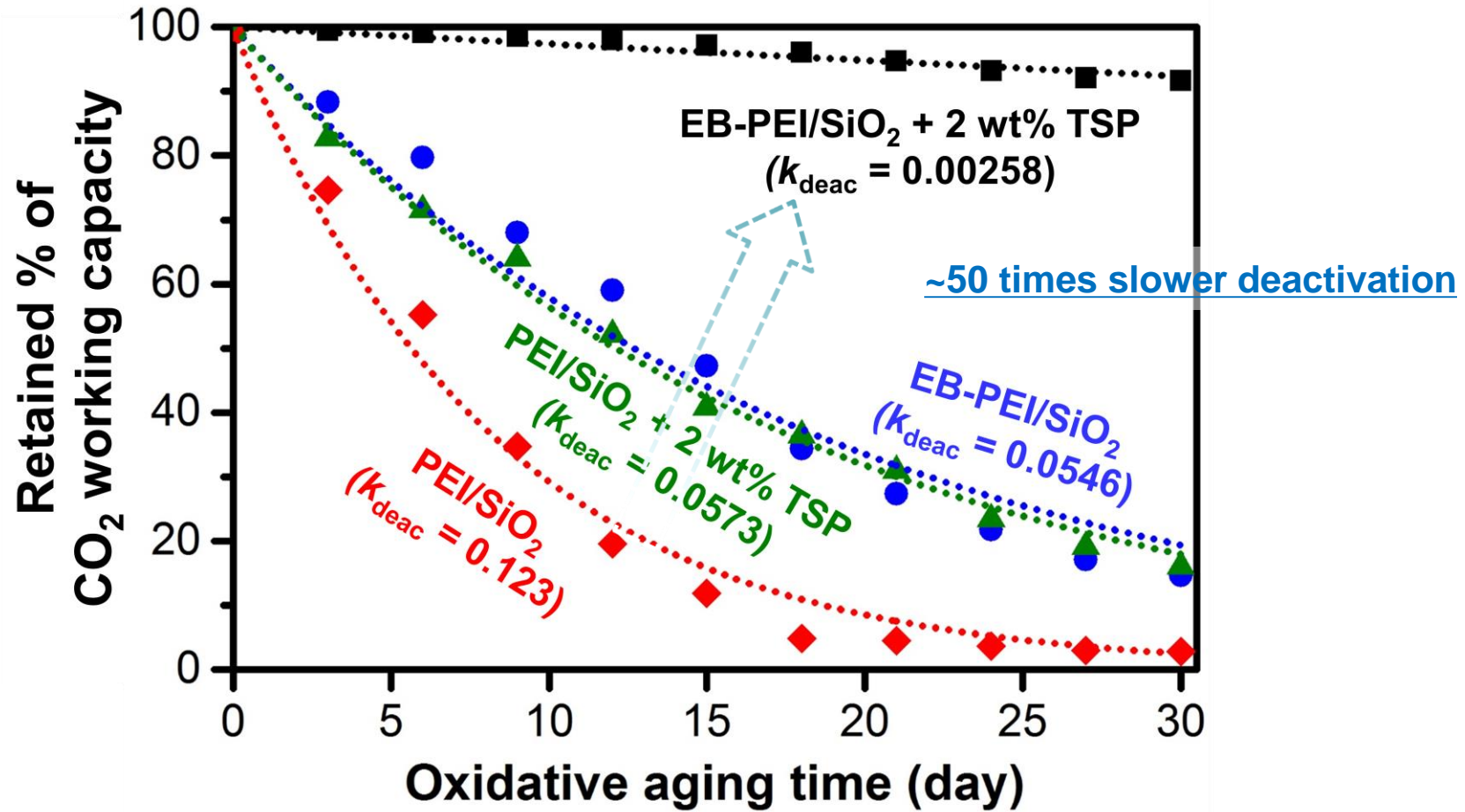
Sample	Heat of adsorption [GJ / ton-CO ₂]
PEI	1.85
0.18EB-PEI	1.73
0.35EB-PEI	1.47
0.49EB-PEI	1.25

▲ ppm-level metal impurity (catalyst for oxidation)

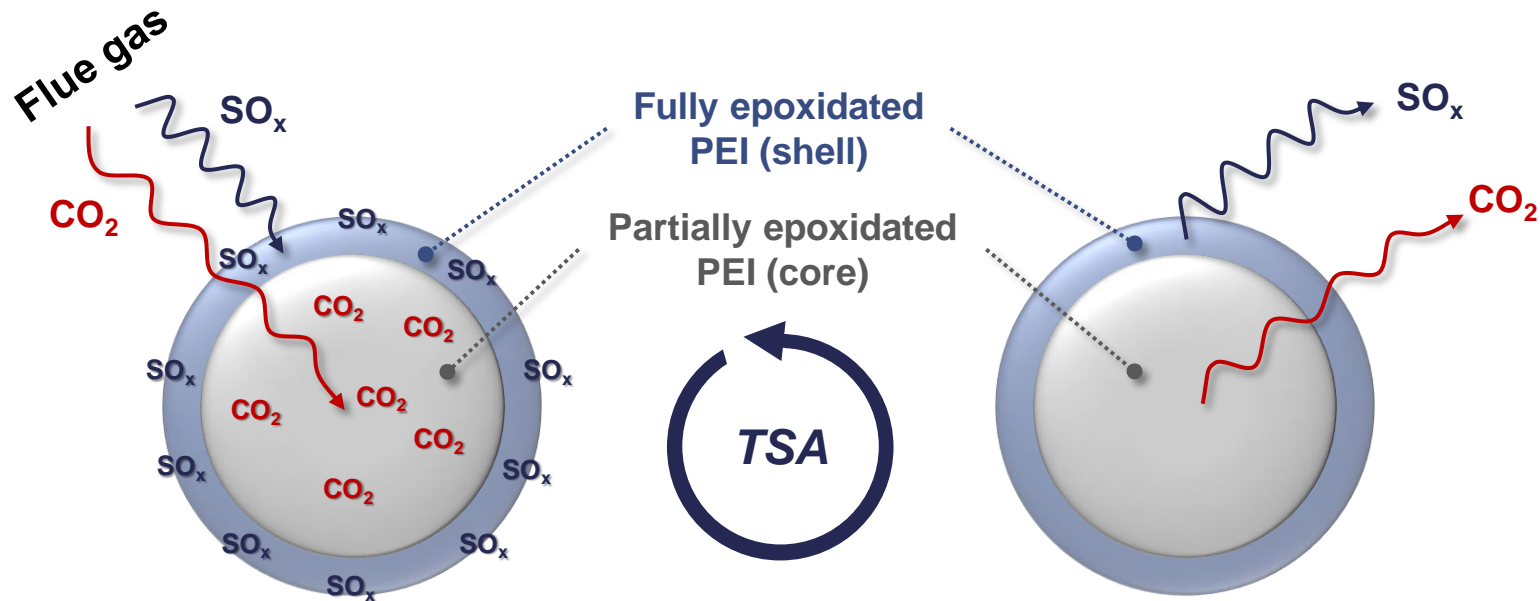


- Solid sorbents meet hot O₂-containing gas in a fluidized bed (standpipe & heat exchanger)
 - O₂ stability is very critical to successful operation of dry sorbent fluidized bed processes.
- Chelators for metal impurity & hydroxylation of PEI retard oxidation of the sorbent.

- *Nature Communications*, **2018**, *9*, 1-7
- Patents (KR 10-2017-0152380, US 2019-0143299)



Oxidative aging condition (simulated flue gas)
- 15% CO₂, 10% H₂O, 3% O₂, balanced with N₂ at 110°C for 30 day



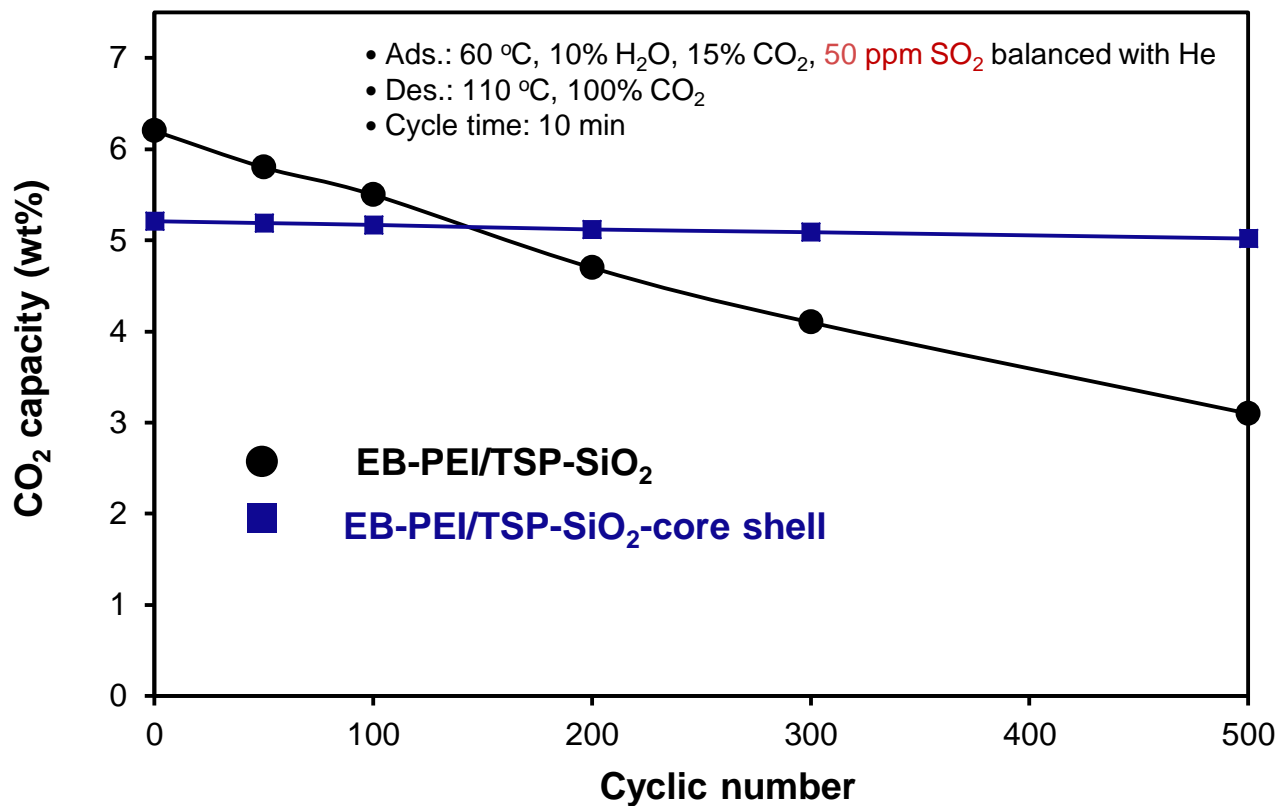
Egg-shell type CO₂ adsorbent

- **Core** : Partially epoxidated PEI

- High CO₂ working capacity
- Highly regenerable without the urea formation
- Irreversible adsorption of SO₂ (SO₂-induced degradation)

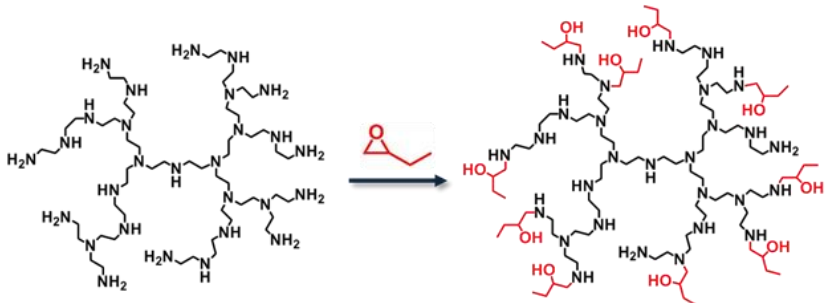
- **Shell** : Fully epoxidated PEI (tertiary amines)

- Selective adsorption of SO₂ over CO₂
- SO₂ can be easily desorbed by increasing temperature (110 °C)
- Protects core-amine (0.37EB-PEI) from the SO₂-induced degradation



* Q_{CO₂} : CO₂ adsorption capacity

Sample	CO ₂ adsorption capacity (wt%)		Q _{CO₂} (SO ₂ treated)/Q _{CO₂} (fresh)*
	Fresh	After 500 cycles	
EB-PEI/TSP-SiO ₂	6.2	3.1	0.50
EB-PEI/TSP-SiO ₂ -core shell	5.2	5.0	0.96

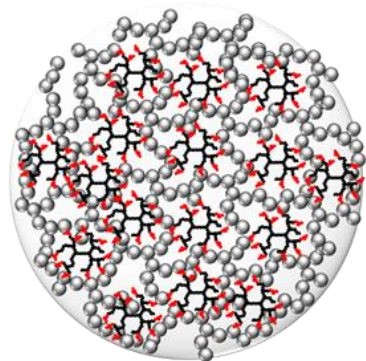


Reaction between epoxide & PEI

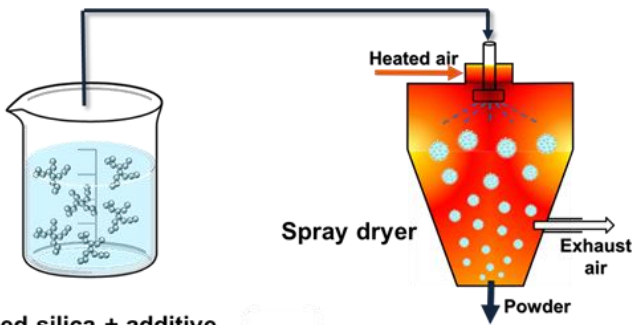


• capable of 500 kg/month manufacture

Impregnation & Drying



Functionalized PEI/silica composite adsorbent



Fumed silica + additive in H₂O solution

Spray dryer



Silica support spray-drying

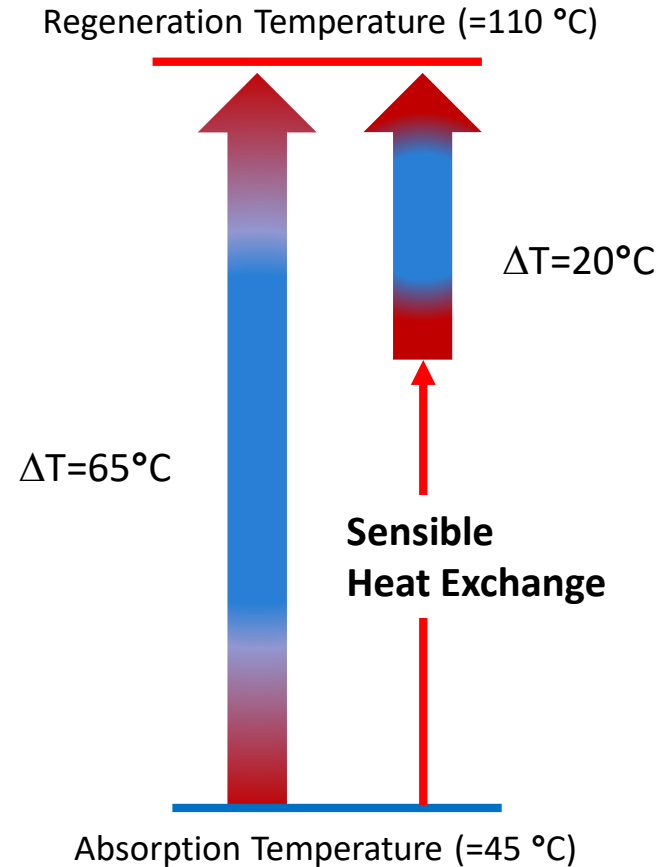
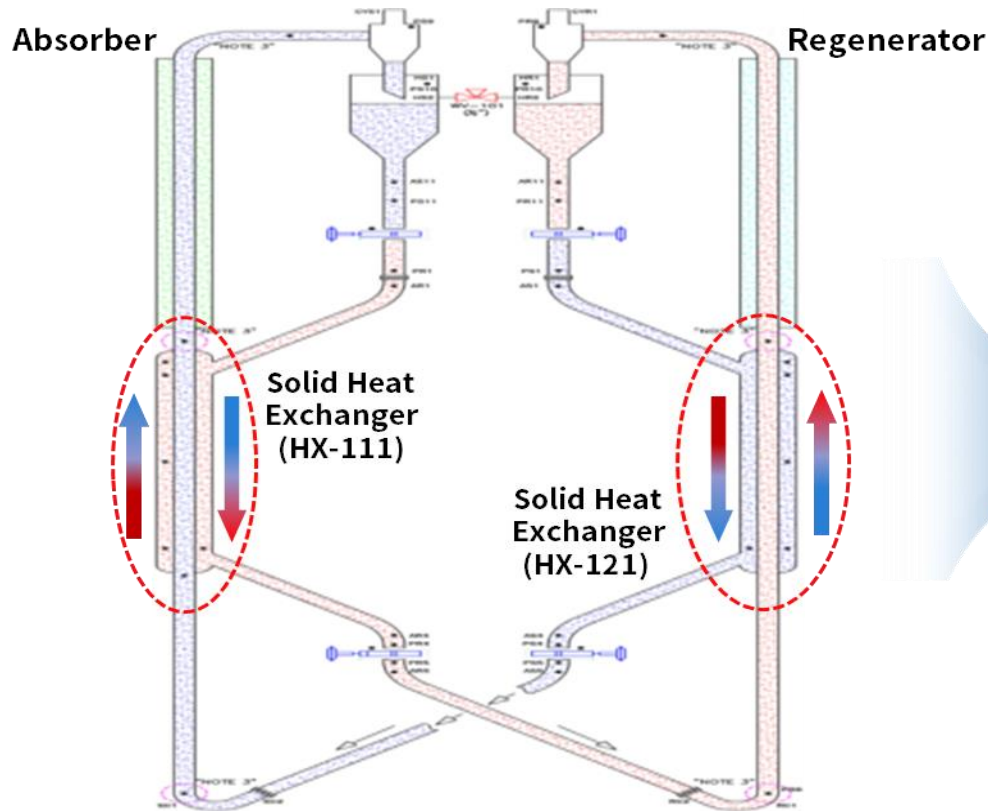
Sorbent properties	
Abs. capacity (wt%)	7.0*
Specific heat (J/g·K)	1.4
Heat of Abs. (GJ/tCO₂)	1.47
Mean particle size (µm)	150
Attrition Index	2.5



*150 mbar CO₂

Concept of solid-solid heat exchanger

100 Nm³/hr bench-scale configuration

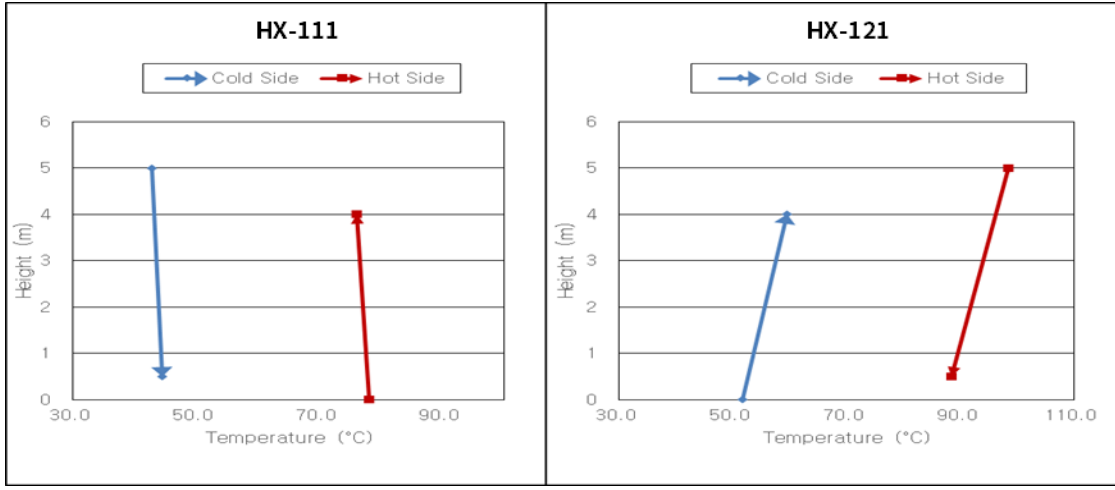
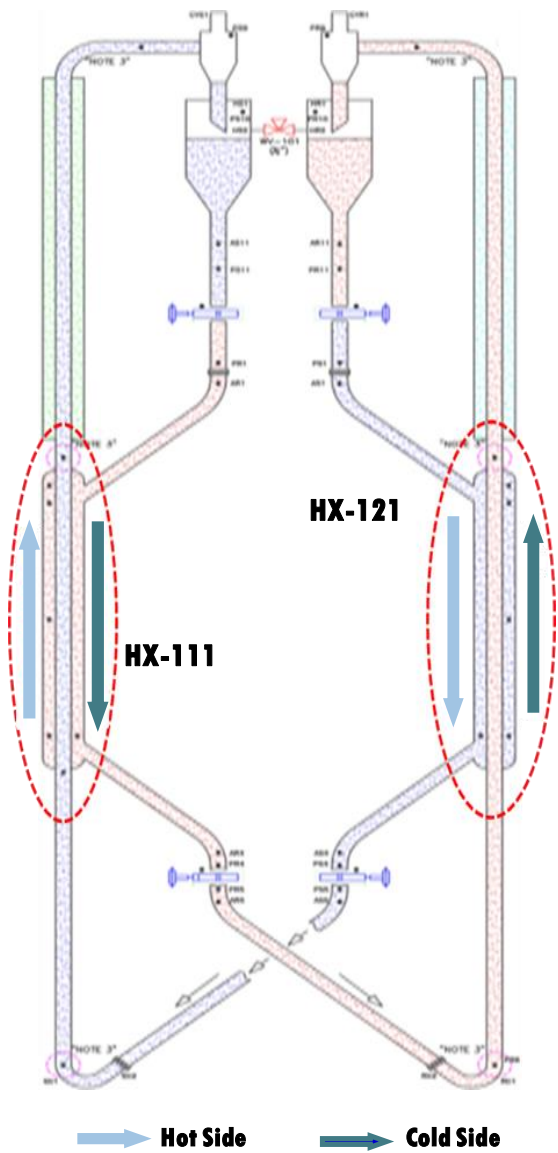


- Patent-pending (EU 14 762 615.4)
- Patent-registered (KR 10-1571966, US 9,694,312)

Recover sensible energy through direct solid-solid heat exchange

($\Delta T: 65 \rightarrow 20^\circ\text{C}$)

Solid-Solid Heat Exchanger Temperature Profile

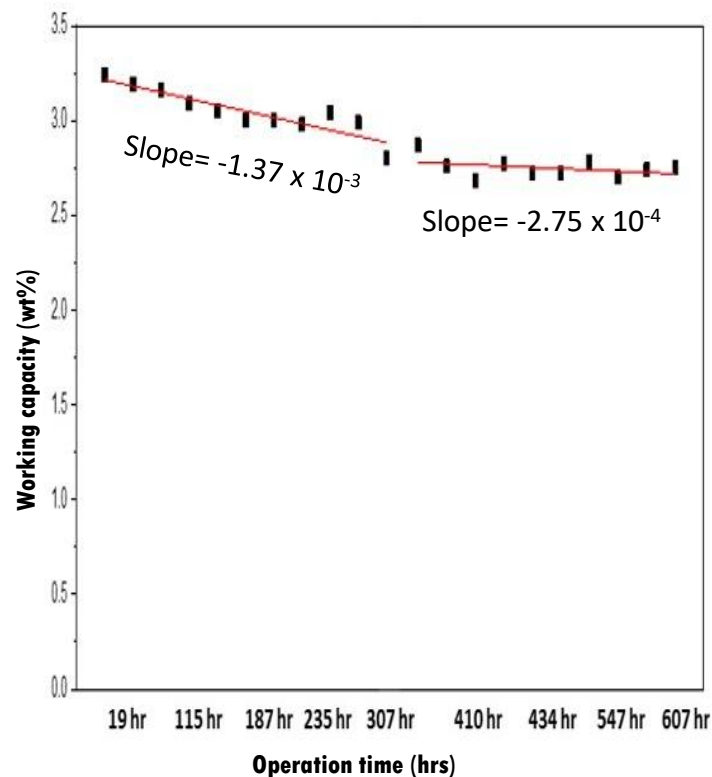
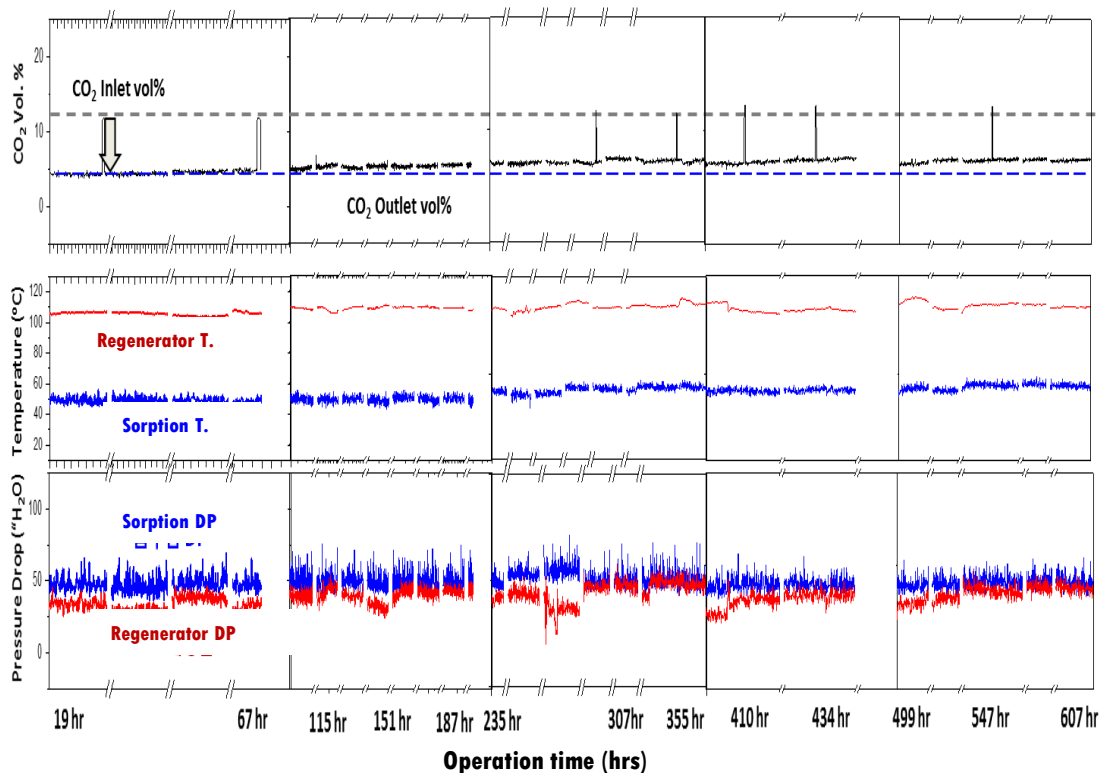


	w/o Solid HX	w/ Solid HX	0.5 MW (Target)
MTA (°C)	65	45	30
Sensible E. (GJ/tCO ₂)	2.6	1.9	1.3

- 20 °C recovered by solid heat exchanger
 → 30% sensible energy reduction
- MTA (minimum temp. approach): 45 °C

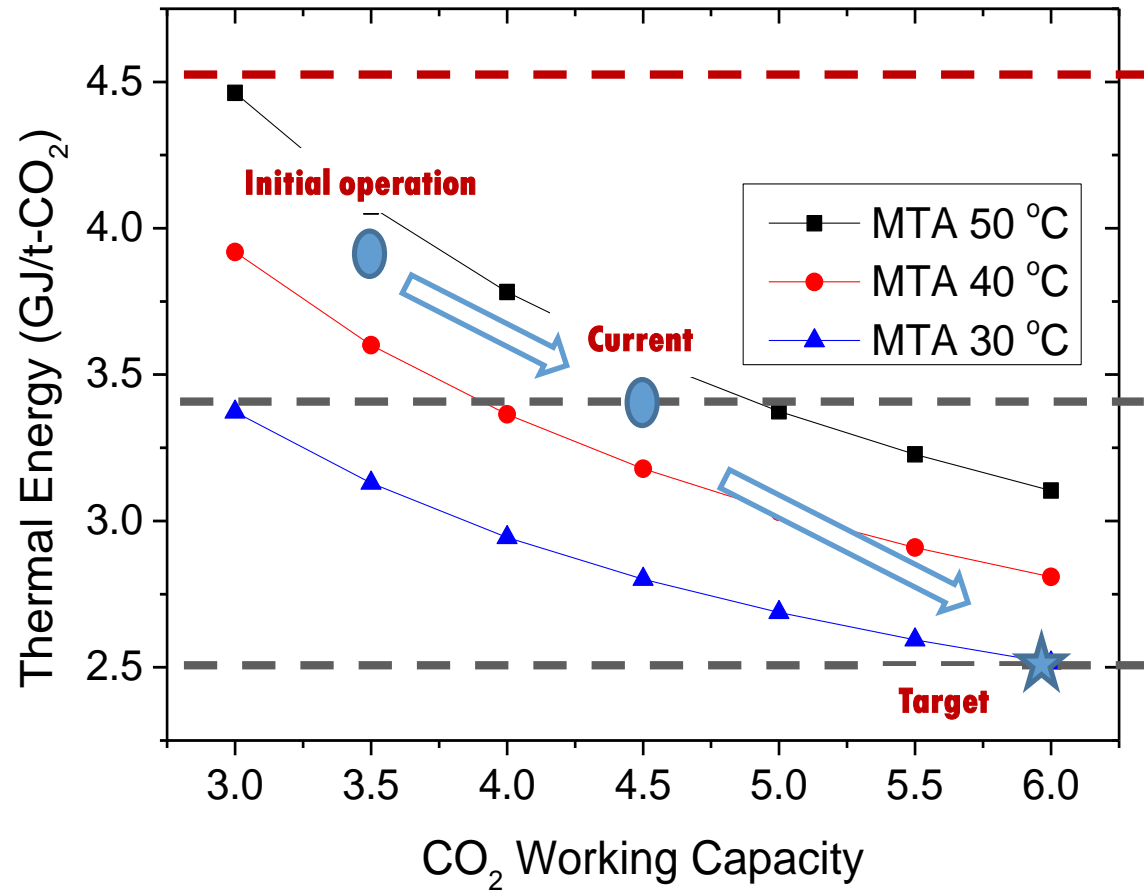
Sorbent stability test and cost

• Bench-scale operation to test stability

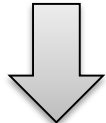


Solvent Type	MEA (Fluor)	KS-1 (MHI)	Benfield (UOP)	EB-PEI
Make up (kg/t-CO ₂)	1.5	0.35 ~ 0.40	2.4	*1.14 (0.5% make-up/day)
Cost (USD/kg)	1.2	16.5	-	12 (raw material cost)

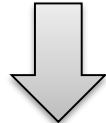
*Basis: attrition loss 0.5wt%/day, sorption working capacity 5wt%



Reference Solid-sorbent Process

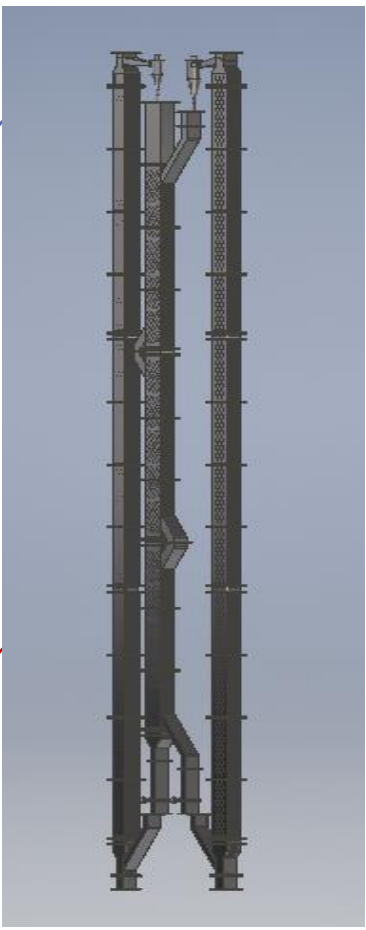
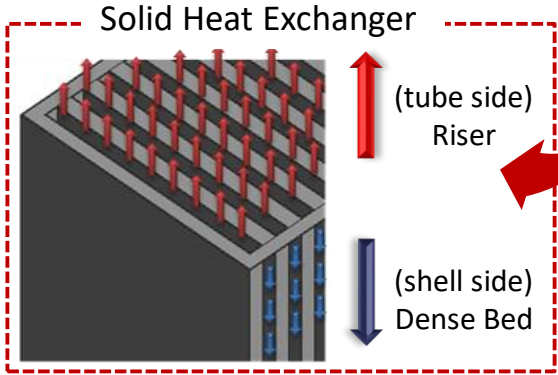
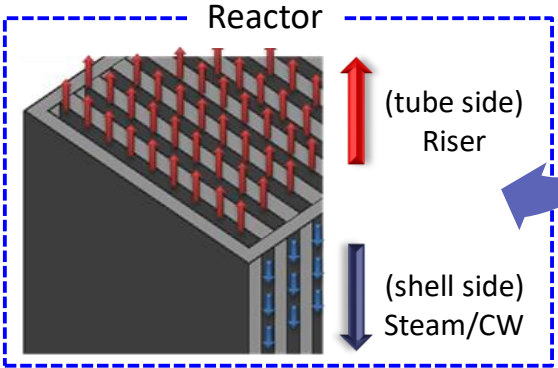


(3.4 GJ/t-CO₂)
100 Nm³/hr



(2.5 GJ/t-CO₂)
0.5 MW Pilot

- Improving MTA by -10 °C - - - - -> Energy reduced by 0.7 GJ/t-CO₂
- Increasing working capacity by +1wt% - - - - -> Energy reduced by 0.5 GJ/t-CO₂

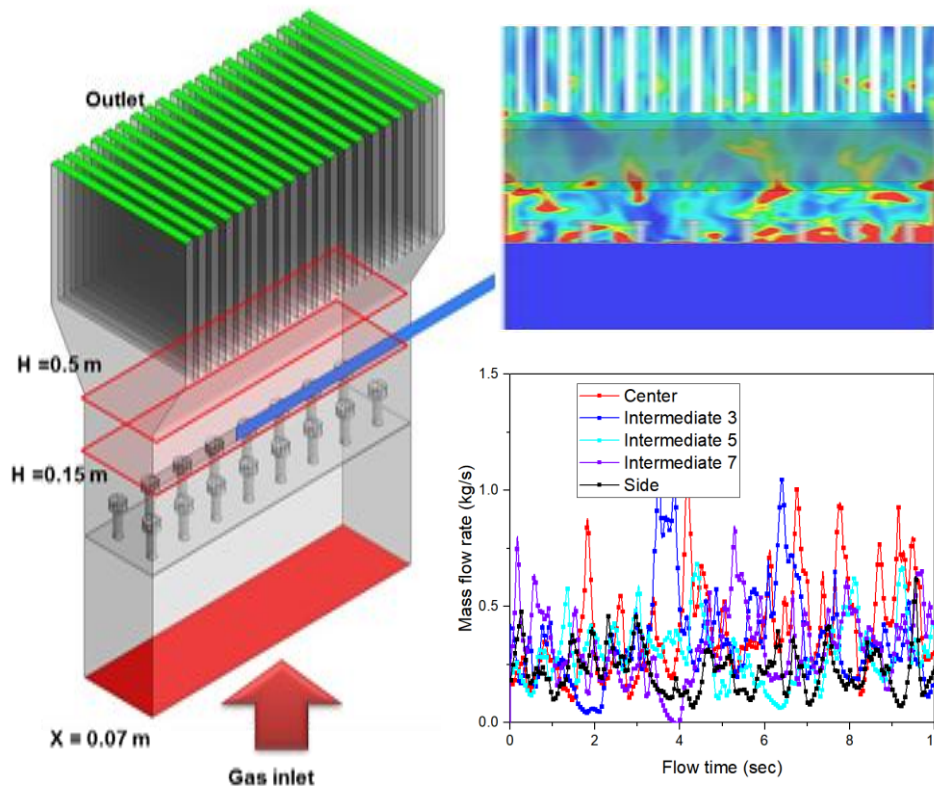


		Absorber	Desorber	Units
Reactor	Heat duty	417	360	kW
	Overall U	35.6	65.1	W/m ² ·K
	Designed Area	613	452	m ²
Solid HX	Heat duty	127	113	kW
	Overall U	15	15	W/m ² ·K
	Designed Area	212	156	m ²
Plate Size	Depth	25	10	mm
	Width	570	570	mm
	Number of Plates	20	15	EA
Footprint		1.15 × 0.63	0.64 × 0.35	m ³
		63 × 35	63 × 35	

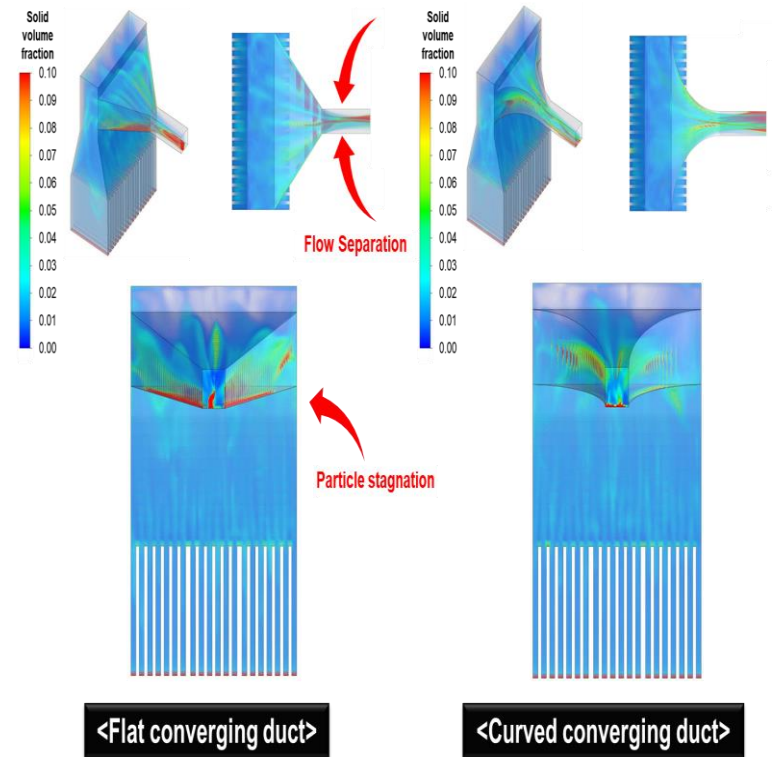
• Reactor : plate-type fluidized bed → easy scale-up by modulation

0.5 MWe pilot plant – CFD analysis

Distributor



Reactor Outlet



- Refer to distributor in CFBC (equipped with Plate-type Super Heater)
- Distributor and reactor outlet designs are under tuned through CFD simulation

Current status of the process development



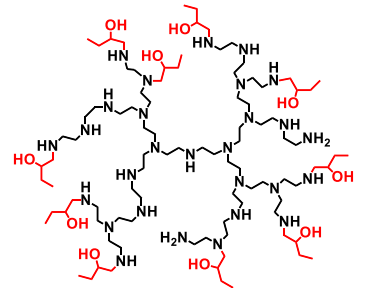
20 Nm³/hr scale operation with restricted heat exchange



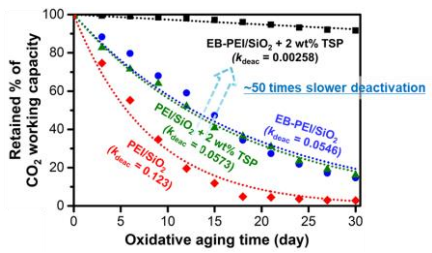
100 Nm³/hr scale operation with improved heat exchange



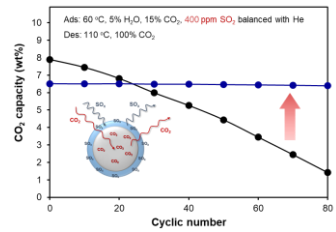
0.5MW pilot test with real flue gas (Target : 2.5 GJ/tCO₂)



EB-PEI



Enhanced O₂ stability



Enhanced SO₂ stability

current

Manufacture of 6 tons of sorbent for pilot test

- A circulating fluidized bed process with a novel solid heat exchanger was developed and it shows 30 % of sensible heat recovery from lean sorbent stream.
- Amine impregnated sorbent with a low heat of sorption was developed and optimized to prevent from thermal and oxidative degradation.
- 1 month continuous operation demonstrates its stability.
- Bench scale process of 20 Nm³/hr and 100 Nm³/hr were constructed and operated.
- Its performance shows 3.4 GJ/t-CO₂ and 4.5 wt% of sorption working capacity.
- 2,000 Nm³/hr (0.5 MW scale) CO₂ capture plant is under constructed and the startup will be from October 2019.

Reactor development

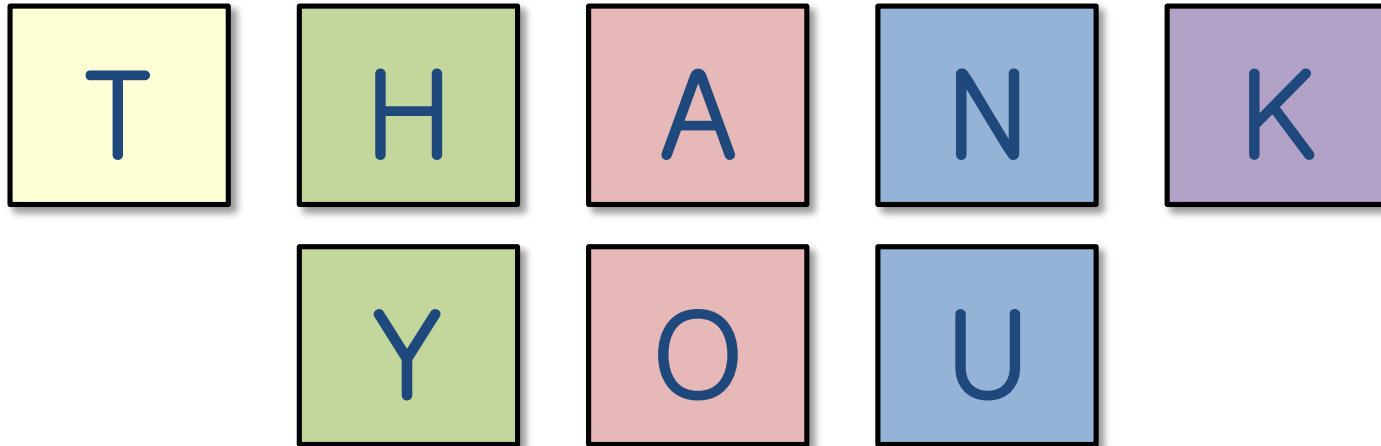
Prof. Hyung Hee Cho (Yeonsei Univ.)

Amine based sorbent development

Prof. Minkee Choi (KAIST)

Prof. Young Soo Ko (Kongju Natl. Univ.)

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Questions & Comments ?